



QUIZ: INTRODUCTION TO MATTER

1. Fill in the blanks with the terms or words that best complete each statement. (5 marks)

- a) As particles of a substance get further apart, the attractive forces between them get weaker.
- b) The temperature at which a liquid becomes a solid is called the freezing point.
- c) The condensing point is the temperature at which gas gas becomes gas liquid.
- d) The term indefinite means that something is changeable.

2. Describe the four postulates of "The Particle Theory of Matter". (4 marks)

- i) All matter is made up of tiny objects called particles.
- ii) All particles have spaces between them.
- iii) All particles are in motion.
- iv) All particles have attractive forces between them.

3. Match each of the following descriptions to the term that best describes it. Each term may be used once, more than once, or not at all. (9 marks)

- | | | |
|-----|---|-----------------|
| I | a) A solid turns into a gas through this process. | A. condensation |
| F | b) This has a definite volume but an indefinite shape. | B. deposition |
| F | c) A _____ turns into a solid through the process of freezing. | C. evaporation |
| Q/D | d) The _____ point of water is 0°C. | D. freezing |
| B | e) A gas turns into a solid through this process. | E. gas |
| C | f) When you boil water, this process is occurring. | F. liquid |
| E/B | g) These two will take the shape of whatever container they are in. | G. melting |
| F | | H. solid |
| A | h) A gas turns into a liquid through this process. | I. sublimation |

4. Describe TWO differences between solids and gases. (4 marks)

Any two of the following:

- R1: A solid has a definite shape, whereas a gas has an indefinite shape.
- R2: A solid has a definite volume, whereas a gas has an indefinite volume.
- R3: The spaces between the particles of a solid are smaller than between those of a gas.
- R4: The particles of a solid have less energy than the particles of a gas.
- R5: The particles of a solid move much slower than the particles of a gas.

5. Explain why each of the following statements is FALSE and then correct it. "Liquids have the most energetic particles with the most space between them." (2 marks)

It is false because gases are the ones which have the most energetic particles with the most space between them. Liquids have particle spaces and particle energy in between those of solids and liquids.

Introduction To Matter Answer Key Person

Jessica J Manson



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Chemistry: The Central Science Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Darius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation

Introductory Matter Physics Francesco Simoni, 2018-08-03 This book aims to introduce the reader to basic concepts concerning matter physics describing how fundamental properties of atoms molecules and condensed matter are affected by properties of electrons and by their interaction with electromagnetic waves As an introductory text on basic properties of matter the contents are designed for undergraduate students in electrical engineering It is based on the lectures given by the author for over a decade on Matter Physics and Solid State Physics It focuses on electronic properties to discuss the structure electrical and optical properties of matter and is organized into six chapters The first chapter is a short review of the basic properties of electromagnetic waves giving the basic concepts related to wave propagation to be handled easily to understand the subsequent topics The next chapter on quantum mechanics helps to understand the quantum properties of matter using the simplest formalizations Chapter 3 introduces the core of the book by using quantum mechanics to describe the electronic properties of the atom Then after atomic bonding molecules and condensed matter are discussed before approaching the structural properties of crystal and soft matter The following chapters 4 and 5 are then devoted to electrical properties and optical properties and address the main topics related to solid state and semiconductor physics as well as light matter interaction The final chapter 6 deals with the basic properties of lasers due to the relevance of light sources in everyday life and their widespread use in all branches of engineering

Deception at Work Michael J. Comer, Timothy E. Stephens, 2017-05-15 Deception at Work tells you how to recognize and deal with lies in meetings negotiations discussions and in writing It is guaranteed to make you a more effective and confident operator no matter what job you do Simply leaving the book on your desk for others to see will improve your chances of not being deceived The book exposes deception in all of its forms linking the authors 40 years of experience in dealing with fraudsters with the most recent findings on MRI scanning and the human brain It explains how why and in what circumstances both achievement and exculpatory lies are told and how they can be resolved It sets out a low key but effective plan for dealing with liars in all shapes and sizes from confidence tricksters to malingerers and hard nosed fraudsters This

ground breaking work includes the most comprehensive summary of the clues to deception of any book currently in print

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NCERT Economics Class 12 - For Various State Boards Exams - 1. Introductory Micro Economics 2. Introductory Macro Economic Dr. Harish Chandra Sharma, Dr. Mohan Singhal ,2022-07-14 NCERT ECONOMICS CLASS 12 A BOOK CONTENTS PART A INTRODUCTORY MICRO ECONOMICS UNIT 1 1 ECONOMICS AN INTRODUCTION MICRO ECONOMICS AND CENTRAL PROBLEMS OF AN ECONOMY UNIT 2 2 CONSUMER S EQUILIBRIUM UTILITY ANALYSIS AND INDIFFERENCE CURVE ANALYSIS 3 DEMAND AND LAW OF DEMAND 4 ELASTICITY OF DEMAND UNIT 3 5 PRODUCTION FUNCTION RETURNS TO A FACTOR AND RETURNS TO SCALE 6 COSTS OF PRODUCTION 7 CONCEPTS OF REVENUE 8 PRODUCER S EQUILIBRIUM MEANING AND CONDITIONS 9 SUPPLY LAW OF SUPPLY AND ELASTICITY OF SUPPLY UNIT 4 5 10 DIFFERENT FORMS OF MARKET MEANING AND FEATURES 11 PRICE DETERMINATION UNDER PERFECT COMPETITION EQUILIBRIUM PRICE PART B INTRODUCTORY MACRO ECONOMICS 1 MACRO ECONOMICS MEANING 2 SOME BASIC CONCEPTS OF MACRO ECONOMICS 3 CIRCULAR FLOW OF INCOME 4 CONCEPTS AND AGGREGATES RELATED TO NATIONAL INCOME 5 MEASUREMENT OF NATIONAL INCOME 6 MONEY MEANING EVOLUTION AND FUNCTION 7 COMMERCIAL BANKS AND CREDIT CREATION 8 CENTRAL BANK MEANING AND FUNCTIONS 9 INDIAN BANKING SYSTEM RECENT REFORMS AND ISSUES 10 AGGREGATE DEMAND AND AGGREGATE SUPPLY 11 DETERMINATION OF EQUILIBRIUM LEVEL OF INCOME EMPLOYMENT AND OUTPUT 12 INVESTMENT MULTIPLIER AND ITS WORKING 13 PROBLEMS OF DEFICIENT AND EXCESS DEMAND 14 MEASURES TO CORRECT DEFICIENT EXCESS DEMAND 15 GOVERNMENT BUDGET AND THE ECONOMY 16 FOREIGN EXCHANGE RATE 17 BALANCE OF PAYMENTS

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