

## PROBLEM 2-37

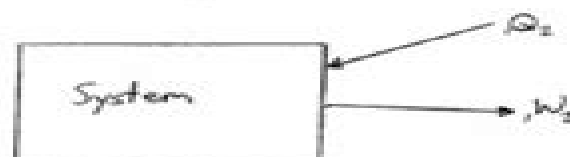
GIVEN: A system of fixed mass at constant pressure and temperature

FIND: Equivalent system form of Eqn. 2.35 (First Law) which is used to define the heat of reaction.

ASSUMPTIONS: expansion/contraction of the system boundary is a reversible process; only work done by system is boundary expansion/contraction

APPROACH: Write the first law for the system. Substitute the appropriate expression for constant pressure work and solve for the specific heat transfer.

Sketch of system:



first law for a system:

$$Q_2 - W_2 = m(u_2 - u_1)$$

for a system at constant pressure the reversible work can be expressed as

$$W_2 = \int P dV = P(V_2 - V_1) = mP(v_2 - v_1) \quad v \equiv \text{specific volume}$$

substituting into the first law

$$Q_2 - mP(v_2 - v_1) = m(u_2 - u_1)$$

$$Q_2 = m[(u_2 + Pv_2) - (u_1 + Pv_1)] = m(h_2 - h_1)$$

$$q_2 = Q_2/m = h_2 - h_1$$

if state 1 is the reactants and state 2 is the products then

$$q = h_p - h_r \quad \leftarrow \text{same form as Eqn 2.35}$$

COMMENTS: Note that the first law for a constant pressure system has the same form as for a control volume. This is only true for constant pressure. Also, the internal energies  $u_1$  and  $u_2$  are mixture internal energies. Even assuming ideal gas behavior,  $u_1 \neq u_2$  at constant temperature if the composition changes. (see Joule's experiment with gases and internal energy)

# Introduction To Combustion Turns Solution Manual

**Rosina Ehmman**



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**Thermodynamics** Stephen R. Turns, Laura L. Pauley, 2020-02-27 Fully revised to match the more traditional sequence of course materials this full color second edition presents the basic principles and methods of thermodynamics using a clear and engaging style and a wealth of end of chapter problems It includes five new chapters on topics such as mixtures psychrometry chemical equilibrium and combustion and discussion of the Second Law of Thermodynamics has been expanded and divided into two chapters allowing instructors to introduce the topic using either the cycle analysis in Chapter 6 or the definition of entropy in Chapter 7 Online ancillaries including new LMS testbanks a password protected solutions manual prepared PowerPoint lecture slides instructional videos and figures in electronic format are available at [www.cambridge.org/thermo](http://www.cambridge.org/thermo)

Transport Phenomena in Fires Mohammad Faghri, Bengt Sundén, 2008 Controlled fires are beneficial for the generation of heat and power while uncontrolled fires like fire incidents and wildfires are detrimental and can cause enormous material damage and human suffering This edited book presents the state of the art of modeling and numerical simulation of the important transport phenomena in fires It describes how computational procedures can be used in analysis and design of fire protection and fire safety Computational fluid dynamics turbulence modeling combustion soot formation thermal radiation modeling are demonstrated and applied to pool fires flame spread wildfires fires in buildings and other examples      *Applied Mechanics Reviews*, 1949      *Forthcoming Books* Rose Army, 1999-08      **A Manual of Inorganic Chemistry**

Anonymous, 2023-03-03 Reprint of the original first published in 1874      A Manual of inorganic chemistry v. 1 Sir Thomas Edward Thorpe, 1873      A Manual of Inorganic Chemistry Thomas Edward Thorpe, 1874      **A manual of inorganic chemistry** sir Thomas Edward Thorpe, 1874      A Manual of Inorganic Chemistry ...: The non-metals Thomas Edward Thorpe, 1874      Technical Manual United States. War Department, 1963      **Introduction to Chemistry, Laboratory**

**Manual** T. R. Dickson, 1994-12-23 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and a chemical vision learning problem solving methods and utilizing group work and discussion activities These themes involve and engage the students in their own learning processes they are challenged to be active The presentation of topics has been altered to include a new chapter which introduces the students to scientific thinking and shows that chemistry involves interesting and relevant topics The reorganization presents many core concepts in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the chapters referring to health technology the environment and society as well as to specific tools of direct use to students      **English Mechanic and Mirror of Science**

,1874     **An Introduction to Combustion Concepts and Applications** Kent M. Van De Graaff, Stephen R. Turns, 1995-12  
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