General Instructions:

Read the following instructions very carefully and strictly follow them:

- (i) This question paper comprises 15 questions. All questions are compulsory.
- (ii) This question paper is divided into three sections A, B and C.
- Section A Questions No. 1 to 7 are short answer type questions. Each question carries 2 marks.
- Section B Questions No. 8 to 13 are also short answer type questions. Each question carries 3 marks.
- Section C Questions No. 14 and 15 are case-based questions. Each question carries 4 marks.
- (vi) Internal choices have been provided in some questions. Only one of the alternatives has to be attempted.

SECTION A

- 1. (a) Write the molecular formula of the following carbon compounds:
 - (i) Methane
 - (ii) Propane
 - (b) Carbon compounds have low melting and boiling points. Why?
- The electrons in the atoms of two elements X and Y are distributed in three shells having 1 and 7 electrons respectively in their outermost shells.
 - (a) Write the group numbers of these elements in the Modern Periodic Table.
 - (b) Write the molecular formula of the compound formed when X and Y combine with each other.
 - (c) Which of the two is electropositive?
- 3. (a) Which of the following flowers will have higher possibility of self-pollination?

Mustard, Papaya, Watermelon, Hibiscus

- (b) List the two reproductive parts of a bisexual flower.
- 4. Which one of the two multicellular organisms Spirogyra and Planaria reproduces by regeneration and why? Give an example of any other organism which can also reproduce by the same process.

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Table of Contents Grade 1 physical Science Question Papers 23

- 1. Understanding the eBook Grade 1 physical Science Question Papers 23
 - The Rise of Digital Reading Grade 1physical Science Question Papers 23
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Grade 1physical Science Question Papers 23
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 1physical Science Question Papers 23
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Grade 1 physical Science Question Papers 23
 - Personalized Recommendations
 - Grade 1physical Science Question Papers 23 User Reviews and Ratings
 - Grade 1physical Science Question Papers 23 and Bestseller Lists

- 5. Accessing Grade 1 physical Science Question Papers 23 Free and Paid eBooks
 - o Grade 1 physical Science Question Papers 23 Public Domain eBooks
 - Grade 1physical Science Question Papers 23 eBook Subscription Services
 - Grade 1physical Science Question Papers 23 Budget-Friendly Options
- 6. Navigating Grade 1physical Science Question Papers 23 eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Grade 1physical Science Question Papers 23 Compatibility with Devices
 - Grade 1physical Science Question Papers 23 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade 1physical Science Question Papers 23
 - Highlighting and Note-Taking Grade 1physical Science Question Papers 23
 - Interactive Elements Grade 1physical Science Question Papers 23
- 8. Staying Engaged with Grade 1physical Science Question Papers 23
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - o Following Authors and Publishers Grade 1physical Science Question Papers 23
- 9. Balancing eBooks and Physical Books Grade 1physical Science Question Papers 23
 - \circ Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 1physical Science Question Papers 23
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade 1 physical Science Question Papers 23
 - Setting Reading Goals Grade 1physical Science Question Papers 23
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade 1 physical Science Question Papers 23
 - Fact-Checking eBook Content of Grade 1physical Science Question Papers 23
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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