

## Matter, Chemical Trends and Chemical Bonding Unit Test

Knowledge	Thinking	Application	Communication
11/14	12/17	13/16	12/12

## Thinking/Investigation (17 marks)

1. An element is composed of 70% of an isotope with an atomic mass of 69.0u. If the other isotope has 2 more neutrons per atom, what is the average atomic mass of the element (4 marks)?

$$AAM = (M_1 \times \text{Abundance}) + (M_2 \times \text{Abundance})$$

$$AAM = (69.0u \times 70\%) + (71.0u \times 30\%)$$

$$AAM = 69.6u$$

4

2. Fill in the following chart (13 marks)

Formula	Lewis Structure	Name(s) of Shape	Overall Polarity of Molecule
$\text{SBr}_2$	$\text{:}\ddot{\text{Br}}\text{--}\ddot{\text{S}}\text{--}\ddot{\text{Br}}\text{:}$ $\text{:}\ddot{\text{Br}}\text{:}$	Trigonal planar ✓	Non-Polar ✓
$\text{PO}_3^{3-}$	$\left[ \text{:}\ddot{\text{O}}\text{--}\ddot{\text{P}}\text{--}\ddot{\text{O}}\text{:} \right]^{-3}$ $\text{:}\ddot{\text{O}}\text{:}$	Trigonal pyramidal ✓	Non-Polar ✗
$\text{HCP}$	$\text{H--}\ddot{\text{C}}\equiv\text{P:}$ why so many ✓	Bent ✗	Non-Polar ✓
$\text{CH}_3\text{NH}_2$	$\begin{array}{c} \text{H} & \text{H} \\   &   \\ \text{H--C--N:} \\   &   \\ \text{H} & \text{H} \end{array}$ ✓	Trigonal planar ✓	Non-Polar ✗

# Grade 11 Chemistry Unit 1 Practice Test

**J Spring**



**Grade 11 Chemistry Unit 1 Practice Test:**

## Whispering the Secrets of Language: An Mental Quest through **Grade 11 Chemistry Unit 1 Practice Test**

In a digitally-driven world wherever displays reign great and instant interaction drowns out the subtleties of language, the profound strategies and mental subtleties concealed within words frequently move unheard. Yet, located within the pages of **Grade 11 Chemistry Unit 1 Practice Test** a captivating fictional treasure sporting with natural thoughts, lies a fantastic quest waiting to be undertaken. Published by a skilled wordsmith, that charming opus encourages visitors on an introspective journey, softly unraveling the veiled truths and profound affect resonating within ab muscles fabric of each and every word. Within the emotional depths with this poignant evaluation, we will embark upon a honest exploration of the book is primary subjects, dissect their interesting publishing type, and yield to the strong resonance it evokes strong within the recesses of readers hearts.

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