

## Lab: Types of Reactions

Reaction: 1 – add zinc to copper sulfate.

|                             |   |
|-----------------------------|---|
| Observations of Reactions   | Solid blue color, texture looks smooth  |
| Predicted types of reaction | I think single replacement,<br>Zn = element > product, reactant<br>CuSO <sub>4</sub> : ionic Compound |
| Observation of products     | Brown solid – copper<br>-Single replacement<br>-zinc replaces copper ion – zinc sulfate               |
| Balanced chemical equations | → Cu(S)+ Zn So <sub>4</sub> (aq)  |
| Types of reactions          | - Single replacement , copper zinc sulfate  |

Reaction: 2 - Mix potassium iodide & lead with nitrate

|                              |  |
|------------------------------|--|
| Observations of reactions    | <ul style="list-style-type: none"><li>- Reactant KI – ionic compound</li><li>- Reactant pb (NO<sub>3</sub>)g – ionic compound</li><li>- Light blue color</li></ul> |
| Predicted types of reactions | Combustion ( and synthesis )   |
| Observation of products      | <ul style="list-style-type: none"><li>- The insoluble yellow us a solid</li></ul>  |

# Grade 10 Types Of Reactions Lab

**Clarice M. Moran, Mary F. Rice**



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**Scientific and Technical Aerospace Reports**, 1982 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database. **Western Electrician**, 1898. *Annual Register* University of Chicago, 1907.

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scientific method boxes that provide students with insight into the relevance of the scientific method to the topic at hand The book also includes in greater depth problems a more challenging probe into certain issues They are more quantitative in nature and require more in depth critical thinking Internet exercises are also integrated throughout the text

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