

Answer following



- 1. Substrate and temperature: 20°C 30°C 40°C
- 2. A graph of product formed: product formed time min
- 3. A graph of substrate concentration: $\text{substrate concentration}$ time min

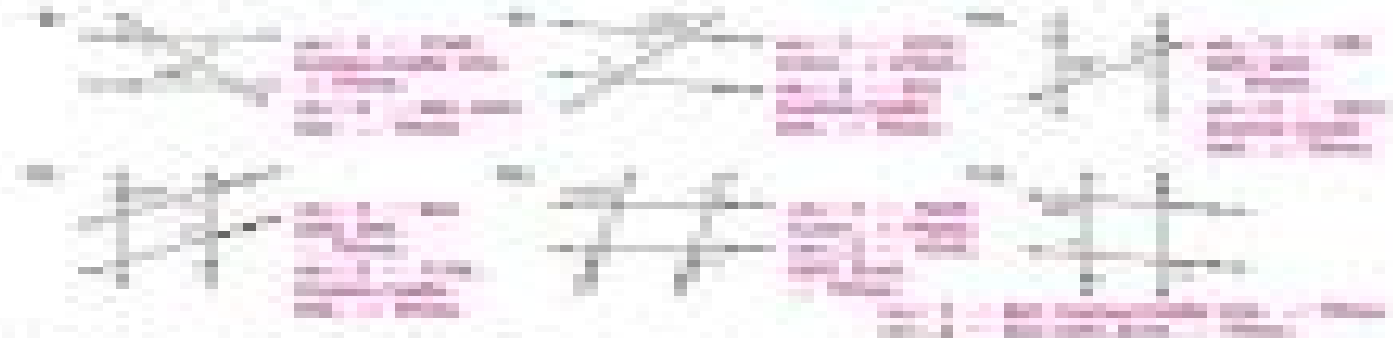
Calculate all values of rate & slope of graph in the diagram.
 (Note: All values are positive and rounded to 2 decimal places)

- 1. maximum rate: 0.001 $\text{mol l}^{-1} \text{s}^{-1}$
- 2. minimum rate: 0.001 $\text{mol l}^{-1} \text{s}^{-1}$
- 3. maximum rate: 0.001 $\text{mol l}^{-1} \text{s}^{-1}$
- 4. minimum rate: 0.001 $\text{mol l}^{-1} \text{s}^{-1}$



Answer 100

Graph 100 is a graph of rate of reaction against time. The graph is shown below.



Graph 100 is a graph of rate of reaction against time. The graph is shown below.



Graph 100 is a graph of rate of reaction against time. The graph is shown below.

Geometry Lesson 10 6 Prentice Hall Answers

J Rink



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