

Unit 1 • Observations, Models, & Experiments

SCIENTIFIC NOTATION

Change the following to Scientific Notation (maintain the number of significant figures):

1. $5.280 = \underline{5.280}$

11. $2,560 = \underline{2.56 \times 10^3}$

2. $2,000 = \underline{2 \times 10^3}$

12. $.0009 = \underline{9 \times 10^{-4}}$

3. $15 = \underline{1.5 \times 10^1}$

13. $8,900,000 = \underline{8.9 \times 10^6}$

4. $6,589,000 = \underline{6.589 \times 10^6}$

14. $.0920 = \underline{9.20 \times 10^{-2}}$

5. $70,400,000,000 = \underline{7.04 \times 10^{10}}$

15. $6,300 = \underline{6.3 \times 10^3}$

6. $.00263 = \underline{2.63 \times 10^{-3}}$

16. $.90 = \underline{9.0 \times 10^{-1}}$

7. $.00589 = \underline{5.89 \times 10^{-3}}$

17. $250 = \underline{2.5 \times 10^2}$

8. $.006 = \underline{6 \times 10^{-3}}$

18. $.006087 = \underline{6.087 \times 10^{-3}}$

9. $.400 = \underline{4.00 \times 10^{-1}}$

19. $500,000 = \underline{5 \times 10^5}$

10. $.08060 = \underline{8.060 \times 10^{-2}}$

20. $.0000000105 = \underline{1.05 \times 10^{-8}}$

Make the following Metric System conversions using "unit analysis" (you may use scientific notation):

1. $100 \text{ mg} \times \frac{1 \text{ g}}{1000 \text{ mg}} = \underline{.1} \text{ g}$

2. $20 \text{ cm} \times \frac{1 \text{ m}}{100 \text{ cm}} = \underline{.2} \text{ m}$

3. $50 \text{ L} \times \frac{1 \text{ kL}}{1000 \text{ L}} = \underline{.05} \text{ kL}$

4. $22 \text{ g} \times \frac{100 \text{ cg}}{1 \text{ g}} = \underline{2200} \text{ cg}$

5. $825 \text{ cm} \times \frac{1 \text{ m}}{100 \text{ cm}} \times \frac{1 \text{ km}}{1000 \text{ m}} = \underline{8.25 \times 10^{-3}} \text{ km}$

6. $2,350 \text{ kg} \times \frac{1000 \text{ g}}{1 \text{ kg}} = \underline{2.35 \times 10^6} \text{ g}$

7. $19 \text{ mL} \times \frac{1 \text{ L}}{1000 \text{ mL}} \times \frac{100 \text{ cL}}{1 \text{ L}} = \underline{1.9} \text{ cL}$

8. $52 \text{ km} \times \frac{1000 \text{ m}}{1 \text{ km}} = \underline{52000} \text{ m}$

9. $36 \text{ m} \times \frac{100 \text{ cm}}{1 \text{ m}} = \underline{3600} \text{ cm}$

10. $18 \text{ cm} \times \frac{1 \text{ m}}{100 \text{ cm}} \times \frac{1000 \text{ mm}}{1 \text{ m}} = \underline{180} \text{ mm}$

11. $6 \text{ g} \times \frac{1000 \text{ mg}}{1 \text{ g}} = \underline{6000} \text{ mg}$

12. $4,259 \text{ mg} \times \frac{1 \text{ g}}{1000 \text{ mg}} = \underline{4.259} \text{ g}$

Estimating Length Using Scientific Notation Answer Key

Rachel Sandford



Estimating Length Using Scientific Notation Answer Key:

Reviewing **Estimating Length Using Scientific Notation Answer Key**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Estimating Length Using Scientific Notation Answer Key**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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