#### Chapter 2 - Essential Chemistry for Biology

(text) - Important

#### Some Basic Chemistry

- Matter: Anything that occupies space nd has mass.
  - Three states: Solid, Liquid, and Gas-
  - Compromised of elements -> substances that cannot be broken down into other substance.
    - All living things share these elements n varying proportion:
      - Sulphor
      - 2) Phosphorus
      - Oxygen.
      - Nitrogen
      - Carbon
         Hydrogen
- 25 essential elements
- Four elements make up about 96% of the weight of most cells:
  - 1) Oscyrgom
  - 2) Carbon
  - Hydrogen
  - 4) Nitrogen
- · Trace Elements: required in very small amounts.

#### Atoms

 Each element consists of one kind of atom -> the smallest unit of matter that still retains its chemical properties.

#### Basic Structure of an Atom

- Atoms are composed of three subatomic particles:
  - Protons (Positive Charge) Found in the nucleus
  - Neutrons (Electrically Neutral) Found in the nucleus
  - Electrons (Negative Charge) Found in the Electron Cloud
- Elements are composed of atoms that differ in the number of subatomic particles.
- · An atom is characterized by:
  - Atomic Number: the number of Protons
  - Mass Numbers: Sum of the number of Protons plus neutrons
    - \* Proton 1 Dalton
    - Neutron 1 Daliton
    - Electron Negligible

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**JS Bruner** 

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