

- 12 A sample of propanone, CH_3COCH_3 , was heated under reflux with potassium dichromate(VI) acidified with sulfuric acid, and then the mixture was distilled. Apart from the peaks due to the $\text{C}-\text{C}$ and $\text{C}-\text{H}$ bonds, what peak(s) would be present in the IR spectrum of the distillate?

- ☐ A A peak due to the $\text{C}=\text{O}$ only.
☐ B A peak due to the $\text{O}-\text{H}$ only.
☐ C Peaks due to $\text{C}=\text{O}$ and $\text{O}-\text{H}$.
☐ D Peaks due to $\text{C}-\text{O}$, $\text{C}=\text{O}$ and $\text{O}-\text{H}$.

(Total for Question 12 = 1 mark)

- 13 Which of the following describes the appearance of iodine under the stated conditions?

	Solid	Dissolved in aqueous potassium iodide	Dissolved in a liquid hydrocarbon
☐ A	purple	brown	purple
☐ B	brown	blue-black	yellow
☐ C	shiny grey	brown	purple
☐ D	shiny grey	brown	brown

(Total for Question 13 = 1 mark)

- 14 Why do calculations of global warming exclude the effect of water vapour in the atmosphere?

- ☐ A Water vapour is not a greenhouse gas.
☐ B Water vapour is a much less potent greenhouse gas than carbon dioxide.
☐ C The average concentration of water vapour in the atmosphere is fairly constant.
☐ D The concentration of water vapour in the atmosphere is much lower than that of carbon dioxide.

(Total for Question 14 = 1 mark)

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 - The Rise of Digital Reading May 2014 Edexcel Chemistry Unofficial Mark Scheme
 - Advantages of eBooks Over Traditional Books
2. Identifying May 2014 Edexcel Chemistry Unofficial Mark Scheme
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an May 2014 Edexcel Chemistry Unofficial Mark Scheme
 - User-Friendly Interface
4. Exploring eBook Recommendations from May 2014 Edexcel Chemistry Unofficial Mark Scheme
 - Personalized Recommendations
 - May 2014 Edexcel Chemistry Unofficial Mark Scheme User Reviews and Ratings
 - May 2014 Edexcel Chemistry Unofficial Mark Scheme and Bestseller Lists

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 - May 2014 Edexcel Chemistry Unofficial Mark Scheme eBook Subscription Services
 - May 2014 Edexcel Chemistry Unofficial Mark Scheme Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - May 2014 Edexcel Chemistry Unofficial Mark Scheme Compatibility with Devices
 - May 2014 Edexcel Chemistry Unofficial Mark Scheme Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of May 2014 Edexcel Chemistry Unofficial Mark Scheme
 - Highlighting and Note-Taking May 2014 Edexcel Chemistry Unofficial Mark Scheme
 - Interactive Elements May 2014 Edexcel Chemistry Unofficial Mark Scheme
8. Staying Engaged with May 2014 Edexcel Chemistry Unofficial Mark Scheme
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers May 2014 Edexcel Chemistry Unofficial Mark Scheme
9. Balancing eBooks and Physical Books May 2014 Edexcel Chemistry Unofficial Mark Scheme
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection May 2014 Edexcel Chemistry Unofficial Mark Scheme
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine May 2014 Edexcel Chemistry Unofficial Mark Scheme
 - Setting Reading Goals May 2014 Edexcel Chemistry Unofficial Mark Scheme
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of May 2014 Edexcel Chemistry Unofficial Mark Scheme
 - Fact-Checking eBook Content of May 2014 Edexcel Chemistry Unofficial Mark Scheme
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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