

3.4

Graph of the change in temperature of a substance as it is heated

(5)



- 3.5 Substance exists as both a solid and a liquid ✓✓ or... is melting (2)
 3.6 37 °C ✓ (2)
 3.7 80 °C ✓ (1)
 3.8 An indication of the average kinetic energy ✓ of the particles in a substance. ✓ (2)
 3.9 Physical ✓ (1)
 3.10 In the solid state the intermolecular forces are the strongest and the particles cannot move. The heater provides thermal energy to the system, which is translated into kinetic energy (increase in temperature) until the intermolecular forces between the particles can be broken. ✓✓ At this point the thermal energy is transformed into latent energy to break the bonds, so no increase in temperature occurs. ✓✓ (4)

[23]

QUESTION 4

- 4.1
 4.1.1 Homogeneous ✓ (1)
 4.1.2 Homogeneous ✓ (1)
 4.1.3 Pure substance ✓ (1)
 4.2
 4.2.1 Potassium bromide ✓ (2)
 4.2.2 Calcium carbonate ✓ (2)
 4.2.3 Ammonium chloride ✓ (2)
 4.3
 4.3.1 ZnSO_4 ✓✓ (2)
 4.3.2 KMnO_4 ✓✓ (2)
 4.3.3 H_2O ✓✓ (2)

[17]

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Table of Contents Grade 1march 2014 Life Science Question Paper

1. Understanding the eBook Grade 1march 2014 Life Science Question Paper
 - The Rise of Digital Reading Grade 1march 2014 Life Science Question Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade 1march 2014 Life Science Question Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 1march 2014 Life Science Question Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 1march 2014 Life Science Question Paper
 - Personalized Recommendations
 - Grade 1march 2014 Life Science Question Paper User Reviews and Ratings
 - Grade 1march 2014 Life Science Question Paper and Bestseller Lists
5. Accessing Grade 1march 2014 Life Science Question Paper Free and Paid eBooks
 - Grade 1march 2014 Life Science Question Paper Public Domain eBooks

- Grade 1march 2014 Life Science Question Paper eBook Subscription Services
- Grade 1march 2014 Life Science Question Paper Budget-Friendly Options
- 6. Navigating Grade 1march 2014 Life Science Question Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade 1march 2014 Life Science Question Paper Compatibility with Devices
 - Grade 1march 2014 Life Science Question Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade 1march 2014 Life Science Question Paper
 - Highlighting and Note-Taking Grade 1march 2014 Life Science Question Paper
 - Interactive Elements Grade 1march 2014 Life Science Question Paper
- 8. Staying Engaged with Grade 1march 2014 Life Science Question Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade 1march 2014 Life Science Question Paper
- 9. Balancing eBooks and Physical Books Grade 1march 2014 Life Science Question Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 1march 2014 Life Science Question Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade 1march 2014 Life Science Question Paper
 - Setting Reading Goals Grade 1march 2014 Life Science Question Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade 1march 2014 Life Science Question Paper
 - Fact-Checking eBook Content of Grade 1march 2014 Life Science Question Paper
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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