

STANZA BOPAPE SECONDARY SCHOOL**GRADE 11 SCOPE FOR PHYSICAL SCIENCES****JUNE EXAMINATIONS****PAPER 1: PHYSICS****DURATION: 2 HOURS****MARKS: 100****TERMINOLOGY**

Vector, Scalar, resultant vector, normal force, frictional force, inertia, newton 1st, 2nd and 3rd laws, Newton's law of universal gravitation, Coulomb's law, Electric field, electric field at a point, ohm's law, Emf, Current strength, power

QUESTION 1 8 multiple choice questions All topics covered in this scope	[16]
QUESTION 2 - Vectors in 2D questions. - Solving problems involving horizontal and vertical components; resultant vectors, closed vector diagrams	[10]
QUESTION 3 - Questions on Newton's law of gravitation - Solve problems on gravitational forces, weight and gravitational acceleration	[8]
QUESTION 4 - Questions on single body systems on the flat surface, incline plane or hanging, with or without friction. - Drawing free-body diagrams - Solving problems by applying Newton's 1st, 2nd or 3rd law	[16]

QUESTION 5 - Questions on two body systems (joined with a string) on the flat surface, incline plane or hanging, with or without friction - Drawing free-body diagrams - Solving problems by applying Newton's 1st, 2nd or 3rd law - Questions on investigating relationships between acceleration, net force and mass given data tables and/or graphs	[16]
QUESTION 6 - Questions on electrostatic forces and electric field between charges - Electric field patterns - Solving problems involving electrostatic forces and electric field for two or more charges	[14]
QUESTION 7 - Questions on Ohm's law experiment including ohmic and non-ohmic conductors - Solving circuit problems with a maximum of four resistors connected in series and parallel combinations - Explain differences in brightness of light bulbs	[18]
TOTAL:	100

PAPER 2: CHEMISTRY**DURATION: 2 HOURS****MARKS: 100****TERMINOLOGY**

Covalent bonding, molecules, valence electrons, dative/coordinate covalent bond, electronegativity, bond energy, bond length, interatomic and intermolecular forces, boiling and melting points, vapour pressure, mole, one mole, molar mass, concentration

QUESTION 1 8 multiple choice questions All topics covered in this scope	[16]
QUESTION 2 - Draw Lewis dot diagrams for atoms, molecules and ions. - Answer questions on molecular shapes, polarity of bonds	[16]
QUESTION 3 - Determine bond polarity and polarity of molecules by using electronegativities - Answer questions on bond energies and bond length given data tables and graphs.	[14]
QUESTION 4 - Identify the type of intermolecular forces given the names or formulae of molecules - Explain differences in boiling points, melting points and vapour pressures by referring to the types of intermolecular forces and/or molecular masses	[16]

QUESTION 5 - Calculations based on molar mass, volume of gas at STP, concentration of aqueous solutions, number of moles, number of atoms/ molecules/ ions etc. - Balance chemical equations	[16]
QUESTION 6 - Determine percentage composition of compounds, empirical and molecular formulae - Solve problems by performing stoichiometric calculations based on balanced equations which include percentage yield.	[20]
TOTAL:	100

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Table of Contents Exam Paper2 2014 June Grade Scope

1. Understanding the eBook Exam Paper2 2014 June Grade Scope
 - The Rise of Digital Reading Exam Paper2 2014 June Grade Scope
 - Advantages of eBooks Over Traditional Books
2. Identifying Exam Paper2 2014 June Grade Scope
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Exam Paper2 2014 June Grade Scope
 - User-Friendly Interface
4. Exploring eBook Recommendations from Exam Paper2 2014 June Grade Scope
 - Personalized Recommendations
 - Exam Paper2 2014 June Grade Scope User Reviews and Ratings
 - Exam Paper2 2014 June Grade Scope and Bestseller Lists
5. Accessing Exam Paper2 2014 June Grade Scope Free and Paid eBooks

- Exam Paper2 2014 June Grade Scope Public Domain eBooks
- Exam Paper2 2014 June Grade Scope eBook Subscription Services
- Exam Paper2 2014 June Grade Scope Budget-Friendly Options
- 6. Navigating Exam Paper2 2014 June Grade Scope eBook Formats
 - ePub, PDF, MOBI, and More
 - Exam Paper2 2014 June Grade Scope Compatibility with Devices
 - Exam Paper2 2014 June Grade Scope Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Exam Paper2 2014 June Grade Scope
 - Highlighting and Note-Taking Exam Paper2 2014 June Grade Scope
 - Interactive Elements Exam Paper2 2014 June Grade Scope
- 8. Staying Engaged with Exam Paper2 2014 June Grade Scope
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Exam Paper2 2014 June Grade Scope
- 9. Balancing eBooks and Physical Books Exam Paper2 2014 June Grade Scope
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Exam Paper2 2014 June Grade Scope
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Exam Paper2 2014 June Grade Scope
 - Setting Reading Goals Exam Paper2 2014 June Grade Scope
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Exam Paper2 2014 June Grade Scope
 - Fact-Checking eBook Content of Exam Paper2 2014 June Grade Scope
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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