

Extended Response Question

On a separate piece of paper answer the following question. (Remember to use the school heading)

Considering the amount of incidences that have happen with Joey, do you think that Mrs. Maxy's treatment of Joey is fair? Why or Why not? Explain using specific examples from Chapter 1-8.

Extended Response Techniques Social Studies

**Murray Shukyn, Dale E.
Shuttleworth, Achim K. Krull**



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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g}) \quad \Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \quad \Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{g}) \quad \Delta h = 283 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g})$

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web thus for reaction 9 2 in which 1 mole of h_3o^+ and 1 mole of oh^- combine to form 2 moles of h_2o $h = 55.8 \text{ kJ}$ as shown note that the enthalpy change $h_{products} - h_{reactants}$ is positive if heat is absorbed that is if $h_{products}$

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reactions 1 naoh s na aq oh aq h 1 2 naoh s h aq cl aq h 2 o na aq cl aq h 2 3

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