

Lab 3: Exploring the Properties of Gases

MONDAY, 9/27/2021

Pre-Lab Grade (15 pts):

Post-Lab Grade (15 pts):

Author:

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Objectives:

The purpose of this experiment would be to investigate the different gas laws involving four gas properties: pressure, volume, temperature, and the number of molecules and their relationships. The experiment will address the question whether changing one gas property would affect the other gas properties based on their relationships.

Procedure:

Part 1: Pressure and Volume

- ☐ 1. Plug the LabQuest, connect the gas pressure sensor to channel one of the interface, and turn it on.
- ☐ 2. Use a 20 mL syringe and set up the front edge of the piston at the 15 mL mark, which indicates the amount of air trapped inside the syringe.
- ☐ 3. Securely connect the syringe to the valve of the gas pressure sensor.
- ☐ 4. Put LabQuest at time based mode, wait 60 seconds for the system to equilibrate and then push the play button and record your data as run 1.
- ☐ 5. Keep the syringe securely connected to the sensor and push the piston of the syringe to the number of 13 mL indicating the decrease of the volume of the air inside the syringe by 2 mL. Record the pressure value for this volume as run 2 into the lab notebook.
- ☐ 6. Redo step 5 with three different volumes at 11, 9, and 7 mL and record the pressure values respectively into the lab notebook.
- ☐ 7. Use the volume and pressure data collected and create a graph.

Part 2: Pressure and Absolute Temperature

- ☐ 1. Insert the temperature probe to channel 2 of the interface. Delete the values from experiment 1.
- ☐ 2. Use 600 mL beaker and 125 mL Erlenmeyer flask and construct pressure and absolute temperature experimental setup. Secure tightly the plastic tubing and the Luer-lock valve to the rubber stopper. Close the Erlenmeyer flask with the rubber stopper tightly. The Luer-lock valve should be put in a closed position.
- ☐ 3. Instill 150 mL of water in the 600 mL beaker and holding the flask, submerge it into the water.
- ☐ 4. Fill the beaker with ice to the level of the flask's neck. Keep the rubber stopper dry. Insert the temperature probe into the beaker. The temperature goal is between 0 and 10 °C. After 60 seconds, when temperature and pressure readings stabilized, record the temperature and pressure values into your notebook.
- ☐ 5. Remove the cold water from the beaker and dry the flask.
- ☐ 6. Instill 300 mL of room temperature water in the 600 mL beaker and holding the flask, submerge it into the water. Keep the rubber stopper dry. Insert the temperature probe into the beaker. The temperature goal is between 20 and 30 °C. After 60 seconds, when temperature and pressure readings stabilized, record the temperature and pressure values into your notebook.
- ☐ 7. Remove the cold water from the beaker and dry the flask.

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Table of Contents Exploring The Properties Of Gases Lab Report

1. Understanding the eBook Exploring The Properties Of Gases Lab Report
 - The Rise of Digital Reading Exploring The Properties Of Gases Lab Report
 - Advantages of eBooks Over Traditional Books
2. Identifying Exploring The Properties Of Gases Lab Report
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Exploring The Properties Of Gases Lab Report
 - User-Friendly Interface
4. Exploring eBook Recommendations from Exploring The Properties Of Gases Lab Report
 - Personalized Recommendations
 - Exploring The Properties Of Gases Lab Report User Reviews and Ratings
 - Exploring The Properties Of Gases Lab Report and Bestseller Lists
5. Accessing Exploring The Properties Of Gases Lab Report Free and Paid eBooks
 - Exploring The Properties Of Gases Lab Report Public Domain eBooks
 - Exploring The Properties Of Gases Lab Report eBook Subscription Services
 - Exploring The Properties Of Gases Lab Report Budget-Friendly Options

6. Navigating Exploring The Properties Of Gases Lab Report eBook Formats
 - ePub, PDF, MOBI, and More
 - Exploring The Properties Of Gases Lab Report Compatibility with Devices
 - Exploring The Properties Of Gases Lab Report Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Exploring The Properties Of Gases Lab Report
 - Highlighting and Note-Taking Exploring The Properties Of Gases Lab Report
 - Interactive Elements Exploring The Properties Of Gases Lab Report
8. Staying Engaged with Exploring The Properties Of Gases Lab Report
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Exploring The Properties Of Gases Lab Report
9. Balancing eBooks and Physical Books Exploring The Properties Of Gases Lab Report
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Exploring The Properties Of Gases Lab Report
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Exploring The Properties Of Gases Lab Report
 - Setting Reading Goals Exploring The Properties Of Gases Lab Report
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Exploring The Properties Of Gases Lab Report
 - Fact-Checking eBook Content of Exploring The Properties Of Gases Lab Report
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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