Name: Ryan Villani Date: 02/15/2017

# Student Exploration: Calorimetry Lab

Vocabulary: calorie, calorimeter, joule, specific heat capacity

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. The Latin word calor means "heat," and meter comes from the Greek word meaning "to

measure." What do you think a calorimeter does? Measures heat.

Where have you heard the word calorie before? What do you think a calorie is? With food, the amount of energy it takes to heat up one ml of water.

#### Gizmo Warm-up

A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up.

Calorimeters can be used to find a substance's **specific heat capacity**. You will use the Calorimetry Lab Gizmo™ to determine the specific heat capacities of various substances.



- On the SIMULATION pane, select Copper. Use the slider to set its Mass to 200 g. Set the Water mass to 200 g. Check that the Water temp is set to 30.0 °C and the copper's Temp is 90 °C. Select the GRAPH tab, and click Play ( ).
  - A. What was the Final temperature of the copper and the water? 34,96C
  - B. How much did the temperature of the copper change? 55.04C
  - C. How much did the temperature of the water change? 4.96C

# **Explore Learning Gizmo Answer Key Calorimetry**

J Rink

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# **Table of Contents Explore Learning Gizmo Answer Key Calorimetry**

- 1. Understanding the eBook Explore Learning Gizmo Answer Key Calorimetry
  - The Rise of Digital Reading Explore Learning Gizmo Answer Key Calorimetry
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Explore Learning Gizmo Answer Key Calorimetry
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Explore Learning Gizmo Answer Key Calorimetry
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Explore Learning Gizmo Answer Key Calorimetry
  - Personalized Recommendations
  - Explore Learning Gizmo Answer Key Calorimetry User Reviews and Ratings
  - Explore Learning Gizmo Answer Key Calorimetry and Bestseller Lists
- 5. Accessing Explore Learning Gizmo Answer Key Calorimetry Free and Paid eBooks
  - Explore Learning Gizmo Answer Key Calorimetry Public Domain eBooks
  - Explore Learning Gizmo Answer Key Calorimetry eBook Subscription Services
  - Explore Learning Gizmo Answer Key Calorimetry Budget-Friendly Options
- 6. Navigating Explore Learning Gizmo Answer Key Calorimetry eBook Formats

- o ePub, PDF, MOBI, and More
- Explore Learning Gizmo Answer Key Calorimetry Compatibility with Devices
- Explore Learning Gizmo Answer Key Calorimetry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Explore Learning Gizmo Answer Key Calorimetry
  - Highlighting and Note-Taking Explore Learning Gizmo Answer Key Calorimetry
  - Interactive Elements Explore Learning Gizmo Answer Key Calorimetry
- 8. Staying Engaged with Explore Learning Gizmo Answer Key Calorimetry
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Explore Learning Gizmo Answer Key Calorimetry
- 9. Balancing eBooks and Physical Books Explore Learning Gizmo Answer Key Calorimetry
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Explore Learning Gizmo Answer Key Calorimetry
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Explore Learning Gizmo Answer Key Calorimetry
  - Setting Reading Goals Explore Learning Gizmo Answer Key Calorimetry
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Explore Learning Gizmo Answer Key Calorimetry
  - Fact-Checking eBook Content of Explore Learning Gizmo Answer Key Calorimetry
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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