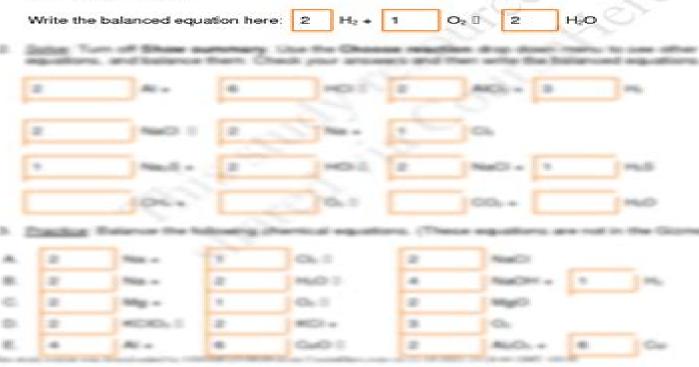
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Activity A:	Get the Gizmo ready:	70.00
Balancing equations	 Check that the Synthesis reaction is selected and that all coefficients are set to one. (The coefficients are the numbers in the boxes.) 	-5

Introduction: The equation $H_2 + O_2 \square H_2O$ is unbalanced because there are two oxygen atoms on the reactants side of the equation, and only one on the products side of the equation. To balance the equation, you cannot change the structure of any of the molecules, but you can change the number of molecules that are used.

Question: How are chemical equations balanced?

 Balance: Turn on Show histograms. The equation is balanced when there are equal numbers of each type of atom represented on each side of the equation. In the Gizmo, use the up and down arrows to adjust the numbers of hydrogen, oxygen, and water molecules until the equation is balanced. When you are done, turn on Show summary to check your answer.



Explore Learning Chemical Equations Answers

David Kirk

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