

# Experiment Problems 1

1. Synthesis reactions:



2. Synthesis by mixing method in dilute solutions:



3. Synthesis solubility product (K<sub>sp</sub>) constants and equilibrium hydrolysis and acids:

|    | Chemical         | Equilibrium                                                               |
|----|------------------|---------------------------------------------------------------------------|
| a) | $\text{Pb}^{2+}$ |                                                                           |
| b) |                  | $2\text{H}^+ + \text{CO}_3^{2-} \rightleftharpoons \text{H}_2\text{CO}_3$ |
| c) | $\text{Ca}^{2+}$ |                                                                           |

|    | Chemical         | Equilibrium             |
|----|------------------|-------------------------|
| a) | $\text{Pb}^{2+}$ |                         |
| b) | $\text{Ca}^{2+}$ |                         |
| c) |                  | $\text{H}_2\text{CO}_3$ |

4. Synthesis equilibrium constants: chemical constants are represented hydrolysis and acids:



# Exponent Practice 1 Unit 09 Lesson 01

**Didier Musso**



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