

EXTRA PRACTICE PROBLEMS FOR MODULE #6

1. Nitric acid is manufactured with the following reaction:



How many grams of nitrogen dioxide are required to produce 5.89×10^3 kg HNO_3 in excess water?

2. Tungsten metal, W, is produced by the following reaction:



How many grams of tungsten can be made from 1.0×10^6 g of hydrogen in excess WO_3 ?

3. How many grams of WO_3 would be used in the above reaction under the conditions specified in #2?

4. Carbon tetrachloride is made according to the following reaction:



How many grams of CS_2 are needed to make 16 kg of CCl_4 ?

5. To make a phosphorus fertilizer ($\text{CaH}_4\text{P}_2\text{O}_8$), agricultural companies use the following reaction:



If 2.50×10^3 grams of $\text{Ca}_3\text{P}_2\text{O}_8$ are reacted with excess H_2SO_4 and H_2O , how many grams of fertilizer can be made?

6. In the following reaction:



What mass of Mg must you use to make 5.6×10^4 g of Mg(OH)_2 ?

7. Which of the following are empirical formulas?

a. SO_3

b. $\text{K}_2\text{S}_2\text{O}_3$

c. $\text{C}_6\text{H}_{12}\text{O}_6$

d. P_2H_4

8. Decomposition of 100.0 g of a dark green powder yields 39.7 g K, 27.9 g Mn, and an unknown amount of oxygen. What is the empirical formula?

9. The mass of the molecule above is 197.1 amu. What is the molecular formula?

10. A compound has an empirical formula of CH_2O . If its molecular mass is 60.0 amu, what is its molecular formula?

Exploring Creation With Chemistry Practice Problem Answers

**E. P. Velikhov, J. M. Gvishiani, S. R.
Mikulinsky**



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Advanced Chemistry in Creation 2nd Edition Solutions and Tests Manual Jay L. Wile,2010-04-01 Solutions and tests manual for Advanced Chemistry in Creation 2nd Edition **Exploring Creation with Chemistry** Jay L. Wile,1996-08 Science, Technology and the Future E. P. Velikhov, J. M. Gvishiani, S. R. Mikulinsky,2013-10-22 Science, Technology and the Future is an analysis of the problems and prospects for the development of science and technology and their role in society. Drawing on the perspectives of Soviet scientists, this book examines the relation between society and nature as well as the prospects for resolving ecological problems with the aid of science and technology. This book is comprised of 33 chapters and begins with a discussion on the role of science and technology in modern society and their place in the solution of global problems. The axiological and ethical aspects of the development of science and the mechanism of scientific and technical progress, economics and social development are also considered. The next section deals with concrete questions pertaining to the development of natural and technical sciences and their significance for the future of mankind, with emphasis on the role of science in the development of productive forces, the state of and the prospects for resolving the energy problem, the most important achievements in the leading branches of physics, chemistry, and biology, opportunities for utilizing space research for man's daily needs, oceanology and geology in the year 2000, science and fertility of the soils, materials for the technology of the future, and prospects for the development of automation and man's place in future production. This monograph will be of interest to sociologists, environmentalists, and science policymakers.

A History of Modern Chemistry Noboru Hirota,2016 This publication is a translation of the book entitled Gendai Kagakushi: A History of Modern Chemistry published by Kyoto University Press in 2013.

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