Name:	Abigail Porter	Date:	

#### Student Exploration: Osmosis

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: cell membrane, concentration, diffusion, dynamic equilibrium, osmosis, semipermeable membrane, solute, solvent

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

 Suppose you were trapped on a desert island with no sources of fresh water. Should you drink water from the ocean? Explain why or why not.

you're gonna die from loss of water, so yes but not a lot. Saltwater is not as safe as freshwater, so by drinking too much saltwater you can mess yourself up.

2. What do you think would happen if you watered your houseplants with salt water?

They would all slowly die.

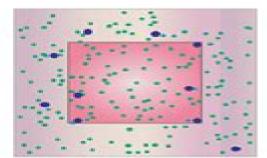
#### Gizmo Warm-up

A cell membrane is a thin "skin" that surrounds a cell. It is a semipermeable membrane, which means that some particles pass through the membrane easily while others cannot.

The Osmosis Gizmo portrays a cell (red square) in a solution of purple solute particles dissolved in green solvent particles. Press Play ( ) and observe.

1. Which particles can pass through the cell membrane?

the green particles (the solvent)



Which particles cannot pass through the cell membrane? the blue

the blue particles (the solute)

3. Click Reset ( 29), and then click Play again. What do you notice about the size of the cell?

It started small and expanded.

# **Explore Learning Osmosis Gizmo Answers**

**AW Rasmussen** 

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#### **Table of Contents Explore Learning Osmosis Gizmo Answers**

- 1. Understanding the eBook Explore Learning Osmosis Gizmo Answers
  - The Rise of Digital Reading Explore Learning Osmosis Gizmo Answers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Explore Learning Osmosis Gizmo Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Explore Learning Osmosis Gizmo Answers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Explore Learning Osmosis Gizmo Answers
  - Personalized Recommendations
  - Explore Learning Osmosis Gizmo Answers User Reviews and Ratings
  - Explore Learning Osmosis Gizmo Answers and Bestseller Lists
- 5. Accessing Explore Learning Osmosis Gizmo Answers Free and Paid eBooks
  - Explore Learning Osmosis Gizmo Answers Public Domain eBooks
  - Explore Learning Osmosis Gizmo Answers eBook Subscription Services
  - Explore Learning Osmosis Gizmo Answers Budget-Friendly Options
- 6. Navigating Explore Learning Osmosis Gizmo Answers eBook Formats

- o ePub, PDF, MOBI, and More
- Explore Learning Osmosis Gizmo Answers Compatibility with Devices
- Explore Learning Osmosis Gizmo Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Explore Learning Osmosis Gizmo Answers
  - Highlighting and Note-Taking Explore Learning Osmosis Gizmo Answers
  - Interactive Elements Explore Learning Osmosis Gizmo Answers
- 8. Staying Engaged with Explore Learning Osmosis Gizmo Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Explore Learning Osmosis Gizmo Answers
- 9. Balancing eBooks and Physical Books Explore Learning Osmosis Gizmo Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Explore Learning Osmosis Gizmo Answers
- 10. Overcoming Reading Challenges
  - $\circ\,$  Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Explore Learning Osmosis Gizmo Answers
  - $\circ\,$  Setting Reading Goals Explore Learning Osmosis Gizmo Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Explore Learning Osmosis Gizmo Answers
  - Fact-Checking eBook Content of Explore Learning Osmosis Gizmo Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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