

Name: _____

Date: _____

Student Exploration: Chemical Equations

Vocabulary: Avogadro's number, chemical equation, chemical formula, chemical reaction, coefficient, combination, combustion, conservation of matter, decomposition, double replacement, molar mass, mole, molecule, product, reactant, single replacement, subscript

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. A candle is placed on one pan of a balance, and an equal weight is placed on the other pan.

What would happen if you lit up the candle and waited for a while? _____

2. Suppose the candle was placed in a large, sealed jar that allowed it to burn for several minutes before running out of oxygen. The candle and jar are balanced by an equal weight.

In this situation, what would happen if you lit up the candle and waited? _____

Gizmo Warm-up

Burning is an example of a **chemical reaction**. The law of **conservation of matter** states that no atoms are created or destroyed in a chemical reaction. Therefore, a balanced **chemical equation** will show the same number of each type of atom on each side of the equation.

To set up an equation in the *Chemical Equations Gizmo™*, type the **chemical formulas** into the text boxes of the Gizmo. First, type in "H₂+O₂" in the **Reactants** box and "H₂O" in the **Products** box. This represents the reaction of hydrogen and oxygen gas to form water.

Reactants	→	Products
H ₂ +O ₂	→	H ₂ O

1. Check that the **Visual** display is chosen on each side of the Gizmo, and count the atoms.
 - A. How many hydrogen atoms are on the **Reactants** side? _____ **Products** side? _____
 - B. How many oxygen atoms are on the **Reactants** side? _____ **Products** side? _____
2. Based on what you see, is this equation currently balanced? _____

Explore Learning Unit Conversions Answers

IM Harris



Explore Learning Unit Conversions Answers:

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Table of Contents Explore Learning Unit Conversions Answers

1. Understanding the eBook Explore Learning Unit Conversions Answers
 - The Rise of Digital Reading Explore Learning Unit Conversions Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Explore Learning Unit Conversions Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Explore Learning Unit Conversions Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Explore Learning Unit Conversions Answers

- Personalized Recommendations
 - Explore Learning Unit Conversions Answers User Reviews and Ratings
 - Explore Learning Unit Conversions Answers and Bestseller Lists
5. Accessing Explore Learning Unit Conversions Answers Free and Paid eBooks
 - Explore Learning Unit Conversions Answers Public Domain eBooks
 - Explore Learning Unit Conversions Answers eBook Subscription Services
 - Explore Learning Unit Conversions Answers Budget-Friendly Options
 6. Navigating Explore Learning Unit Conversions Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Explore Learning Unit Conversions Answers Compatibility with Devices
 - Explore Learning Unit Conversions Answers Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Explore Learning Unit Conversions Answers
 - Highlighting and Note-Taking Explore Learning Unit Conversions Answers
 - Interactive Elements Explore Learning Unit Conversions Answers
 8. Staying Engaged with Explore Learning Unit Conversions Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Explore Learning Unit Conversions Answers
 9. Balancing eBooks and Physical Books Explore Learning Unit Conversions Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Explore Learning Unit Conversions Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Explore Learning Unit Conversions Answers
 - Setting Reading Goals Explore Learning Unit Conversions Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Explore Learning Unit Conversions Answers

- Fact-Checking eBook Content of Explore Learning Unit Conversions Answers
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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