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Date: _____

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? Share

2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

Share 2

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? 7

2. Click **Pause** (). Drag an electron from the left atom to the right atom. Click **Play**.

What happens? it orbits both atoms

3. Click **Pause**, drag an electron from the right atom to the left, and then click **Play**.

What happens now? it orbits both atoms

Explore Learning Student Exploration Covalent Bonds Answer Key

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Table of Contents Explore Learning Student Exploration Covalent Bonds Answer Key

1. Understanding the eBook Explore Learning Student Exploration Covalent Bonds Answer Key
 - The Rise of Digital Reading Explore Learning Student Exploration Covalent Bonds Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Explore Learning Student Exploration Covalent Bonds Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Explore Learning Student Exploration Covalent Bonds Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Explore Learning Student Exploration Covalent Bonds Answer Key

- Personalized Recommendations
 - Explore Learning Student Exploration Covalent Bonds Answer Key User Reviews and Ratings
 - Explore Learning Student Exploration Covalent Bonds Answer Key and Bestseller Lists
5. Accessing Explore Learning Student Exploration Covalent Bonds Answer Key Free and Paid eBooks
 - Explore Learning Student Exploration Covalent Bonds Answer Key Public Domain eBooks
 - Explore Learning Student Exploration Covalent Bonds Answer Key eBook Subscription Services
 - Explore Learning Student Exploration Covalent Bonds Answer Key Budget-Friendly Options
 6. Navigating Explore Learning Student Exploration Covalent Bonds Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Explore Learning Student Exploration Covalent Bonds Answer Key Compatibility with Devices
 - Explore Learning Student Exploration Covalent Bonds Answer Key Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Explore Learning Student Exploration Covalent Bonds Answer Key
 - Highlighting and Note-Taking Explore Learning Student Exploration Covalent Bonds Answer Key
 - Interactive Elements Explore Learning Student Exploration Covalent Bonds Answer Key
 8. Staying Engaged with Explore Learning Student Exploration Covalent Bonds Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Explore Learning Student Exploration Covalent Bonds Answer Key
 9. Balancing eBooks and Physical Books Explore Learning Student Exploration Covalent Bonds Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Explore Learning Student Exploration Covalent Bonds Answer Key
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Explore Learning Student Exploration Covalent Bonds Answer Key
 - Setting Reading Goals Explore Learning Student Exploration Covalent Bonds Answer Key
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Explore Learning Student Exploration Covalent Bonds Answer Key

- Fact-Checking eBook Content of Explore Learning Student Exploration Covalent Bonds Answer Key
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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