

(a)

Reactants	Products
H, Cl and HCl	2H, O <sub>2</sub> and H <sub>2</sub> O

(b)

Reactants	Products
HCl and 2H <sub>2</sub> O	H, Cl, 2H and O <sub>2</sub>

(c)

Reactants	Products
H, Cl, 2H and O <sub>2</sub>	HCl and 2H <sub>2</sub> O

(d)

Reactants	Products
2H, O <sub>2</sub> and H <sub>2</sub> O	H, Cl and HCl

Correct Answer: Option C

Q4) A student performs an experiment to form aluminium chloride from aluminium and chlorine. Which options give the chemical equation of the reaction?

- (a)  $\text{Al} + \text{Cl}_2 \rightarrow \text{AlCl}_2$   
 (b)  $2\text{Al} + \text{Cl}_2 \rightarrow 2\text{AlCl}$   
 (c)  $2\text{Al} + 3\text{Cl}_2 \rightarrow 2\text{AlCl}_3$   
 (d)  $3\text{Al} + 3\text{Cl}_2 \rightarrow 3\text{AlCl}_3$

Correct Answer: Option C

Q5. Give the ratio in which hydrogen and oxygen are present in water by volume.

- (a) 1:2  
 (b) 1:1  
 (c) 2:1

# Mcq Bank On Chemical Reaction And Equation

**ML Yell**



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## Unveiling the Power of Verbal Artistry: An Psychological Sojourn through **Mcq Bank On Chemical Reaction And Equation**

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