

Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_

## Skills Worksheet

**Concept Review****Section: Calculating Quantities in Reactions**

Complete each statement below by writing the correct term or phrase.

1. All stoichiometric calculations involving equations use \_\_\_\_\_ ratios.
2. When solving stoichiometric problems, you must \_\_\_\_\_ the equation first.
3. Balanced equations give the \_\_\_\_\_ numbers of moles of substances.
4. \_\_\_\_\_ in chemical equations provide molar ratios that can be used as conversion factors.
5. The conversion factor for converting between mass and moles is the \_\_\_\_\_ of the substance.
6. In making calculations involving \_\_\_\_\_, you must convert volume to mass.
7. To convert from volume to mass, you can use the \_\_\_\_\_ of the substance as the conversion factor.
8. When calculating the number of particles, you can use \_\_\_\_\_ as the conversion factor.

In the blanks at the left, write the letter of the choice that best completes the statement or answers the question. Consider the following problem when answering:

What mass of sulfuric acid is required to neutralize 2.65 g of potassium hydroxide? The products of the reaction are potassium sulfate and water.

- \_\_\_\_\_ 9. What should you do first after reading the problem carefully?
- a. Estimate the answer.
  - b. Calculate the molar mass of sulfuric acid.
  - c. Write a balanced chemical equation.
  - d. Convert all masses to moles.
- \_\_\_\_\_ 10. What should you do before setting up the problem?
- a. Determine the densities.
  - b. Calculate molar masses.
  - c. Convert all masses to moles.
  - d. Estimate the answer.

# Holt Science Spectrum Chemical Reactions Concept Review

**Michael Hoelscher**



## **Holt Science Spectrum Chemical Reactions Concept Review:**

**Science Spectrum** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03      *Forthcoming Books* Rose Army, 2003      **Introduction to Chemical Dependency Counseling** Jerome David Levin, Joseph Culkin, Richard S. Perrotto, 2001 Suitable for use by students preparing to take the Certified Alcoholism and Substance Abuse Counselor CASAC exams this text provides an introduction to alcoholism and drug addiction      **Technical Book Review**, 1965      Technical Book Review Index, 1968      **Progress in Ecological Stoichiometry** Dedmer B. Van de Waal, James J. Elser, Adam C. Martiny, Robert W. Sterner, James B. Cotner, 2018 Ecological stoichiometry concerns the way that the elemental composition of organisms shapes their ecology. It deals with the balance or imbalance of elemental ratios and how that affects organism growth, nutrient cycling, and the interactions with the biotic and abiotic worlds. The elemental composition of organisms is a set of constraints through which all the Earth's biogeochemical cycles must pass. All organisms consume nutrients and acquire compounds from the environment proportional to their needs. Organismal elemental needs are determined in turn by the energy required to live and grow, the physical and chemical constraints of their environment, and their requirements for relatively large polymeric biomolecules such as RNA, DNA, lipids, and proteins, as well as for structural needs including stems, bones, shells, etc. These materials together constitute most of the biomass of living organisms. Although there may be little variability in elemental ratios of many of these biomolecules, changing the proportions of different biomolecules can have important effects on organismal elemental composition. Consequently, the variation in elemental composition both within and across organisms can be tremendous, which has important implications for Earth's biogeochemical cycles. It has been over a decade since the publication of Sterner and Elser's book *Ecological Stoichiometry* (2002). In the intervening years, hundreds of papers on stoichiometric topics ranging from evolution and regulation of nutrient content in organisms to the role of stoichiometry in populations, communities, ecosystems, and global biogeochemical dynamics have been published. Here we present a collection of contributions from the broad scientific community to highlight recent insights in the field of Ecological Stoichiometry.      **Current Index to Journals in Education**, 1999      *Cumulated Index Medicus*, 1971      **National Library of Medicine Current Catalog** National Library of Medicine (U.S.), 1974 First multi-year cumulation covers six years 1965-70      **Fossil Energy Update**, 1985      **The International Journal for Biosocial Research**, 1982      **Holt Science Spectrum** Ken Dobson, Michael Roberts, 2001      NASA SP., 1962      **Te HS&T J** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004-02      **British Education Index**, 1973      **Publisher and Bookseller**, 1966 Vols for 1871-76, 1913-14 include an extra number. The Christmas bookseller separately paged and not included in the consecutive numbering of the regular series      **The Education Index**, 1971      **Science Citation Index**, 1994 Vols for 1964 have guides and journal lists      **Weed Technology**, 1991      *Paperbacks in Print*, 1971

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