

## Skills Worksheet

## Concept Review Chapter 5: Section 1

## Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

# Holt Modern Chemistry Homework Answers Reviewing Concepts

**Lei Huang**



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## **Table of Contents Holt Modern Chemistry Homework Answers Reviewing Concepts**

1. Understanding the eBook Holt Modern Chemistry Homework Answers Reviewing Concepts
  - The Rise of Digital Reading Holt Modern Chemistry Homework Answers Reviewing Concepts
  - Advantages of eBooks Over Traditional Books
2. Identifying Holt Modern Chemistry Homework Answers Reviewing Concepts
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Holt Modern Chemistry Homework Answers Reviewing Concepts
  - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Modern Chemistry Homework Answers Reviewing Concepts
  - Personalized Recommendations

- Holt Modern Chemistry Homework Answers Reviewing Concepts User Reviews and Ratings
- Holt Modern Chemistry Homework Answers Reviewing Concepts and Bestseller Lists
- 5. Accessing Holt Modern Chemistry Homework Answers Reviewing Concepts Free and Paid eBooks
  - Holt Modern Chemistry Homework Answers Reviewing Concepts Public Domain eBooks
  - Holt Modern Chemistry Homework Answers Reviewing Concepts eBook Subscription Services
  - Holt Modern Chemistry Homework Answers Reviewing Concepts Budget-Friendly Options
- 6. Navigating Holt Modern Chemistry Homework Answers Reviewing Concepts eBook Formats
  - ePub, PDF, MOBI, and More
  - Holt Modern Chemistry Homework Answers Reviewing Concepts Compatibility with Devices
  - Holt Modern Chemistry Homework Answers Reviewing Concepts Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Holt Modern Chemistry Homework Answers Reviewing Concepts
  - Highlighting and Note-Taking Holt Modern Chemistry Homework Answers Reviewing Concepts
  - Interactive Elements Holt Modern Chemistry Homework Answers Reviewing Concepts
- 8. Staying Engaged with Holt Modern Chemistry Homework Answers Reviewing Concepts
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Holt Modern Chemistry Homework Answers Reviewing Concepts
- 9. Balancing eBooks and Physical Books Holt Modern Chemistry Homework Answers Reviewing Concepts
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Holt Modern Chemistry Homework Answers Reviewing Concepts
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Holt Modern Chemistry Homework Answers Reviewing Concepts
  - Setting Reading Goals Holt Modern Chemistry Homework Answers Reviewing Concepts
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Holt Modern Chemistry Homework Answers Reviewing Concepts
  - Fact-Checking eBook Content of Holt Modern Chemistry Homework Answers Reviewing Concepts

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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