

- (A) sodium carbonate
(C) alum

- (B) Sodium sulphate
(D) Sodium chloride

CHEMISTRY

2012

Time: 2 Hours 40 Minutes

Marks: 68

SECTION 'B' (SHORT-ANSWER QUESTIONS)

NOTE: Attempt any 5 questions from Inorganic & 5 questions from Organic Chemistry. (40)

INORGANIC CHEMISTRY

- 2.(i) State the Modern periodic law. Explain how it removed the defects of Mendeleev's System of classification.
- (ii) (a) Write the Valence shell configuration of the following groups. * IA and IB * VA and VB
(b) What is Aqua Regia? How does gold dissolve in it? Give equations for the reaction.
- (iii) Give the reason of the following:
* Ionic Hydrides are called true Hydrides.
* Lithium and Beryllium markedly differ from other members of their respective groups.
* Plaster of Paris is used in making plaster coats and moulds.
* Why is the electronic configuration of Chromium (Cr) $4s^1, 3d^5$ instead of $4s^2, 3d^4$ while that of Copper (Cu) is $4s^1, 3d^{10}$ instead of $4s^2, 3d^9$?
- (iv) Give equations for the following:
* Zinc is treated with 90% conc H_2SO_4
* Action of Heat on $KMnO_4$
* Aluminum reacts with aqueous Sodium Hydroxide
* Lunar caustic is heated $450^\circ C$
- (v) (a) Write a note on any one: Detergents OR Tin plating.
(b) Write I.U.P.A.C. names of the following:
* $[Fe(NO_2)_6]^{3-}$ * $[Cr(en)_2 Cl_2]Cl$
- (vi) Describe the preparation of Chlorine gas by Castner-Kellner cell or Nelson cell.
- (vii) Complete the following equations and give I.U.P.A.C. names of the complexes formed:
* $CuSO_4 + NH_3 \xrightarrow{\text{excess}}$ * $AgNO_3 + NH_3 \xrightarrow{\text{excess}}$
- (viii) (a) What is Blue vitriol? Give two equations for its preparation.
(b) What are outer and inner transition elements and how many series of each of them are present in the periodic table? Also write the names of the inner transition series.

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