

LIFE SCIENCES

Topic test – Grade 10 06.09.2023

Animal Tissues

Marks: 30

Time: 30 minutes

SECTION A

QUESTION 1

1.1 Various options are provided as possible answers to the following questions. Choose the correct answer and write only the letter (A to D) next to the question number (1.1.1 to 1.1.3), for example 1.1.4 D.

1.1.1 Which muscle has a striped appearance?

- A Voluntary muscle
- B Involuntary muscle
- C Cardiac muscle
- D Smooth muscle

1.1.2 The red pigment in blood that carries oxygen is ____

- A found in the leucocytes.
- B found in the platelets.
- C is not found in erythrocytes.
- D known as haemoglobin.

1.1.3 Which of the following are examples of connective tissues?

- (i) Squamous tissue
- (ii) Blood tissue
- (iii) Cuboidal tissue
- (iv) Tendons

- A (i) and (ii) only
- B (ii) and (iii) only
- C (iv) only
- D (ii) and (iv) only

(3 x 2) (6)

1.2 Give the correct biological term for each of the following descriptions.

- 1.2.1 Neurons that transmit nerve impulses from sensory organs to the central nervous system
- 1.2.2 The tissue that gives shape and rigidity to the body
- 1.2.3 Type of epithelial tissue lining the mouth and blood vessels
- 1.2.4 A structure that attaches muscle to bone

(4 x1) (4)

TOTAL SECTION A: [10]

SECTION B

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web conclusion the purpose of the experiment was to identify the unknown solid acid 17 by calculating its molar mass this was done by titration of the acid with sodium hydroxide in order to find the volume of sodium hydroxide required to reach the end point and that value was used in unit analysis to find the molar mass and identify the unknown solid acid

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pka of an unknown acid or pkb of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web this point of titration is referred to as the equivalence point considering the 1:1 stoichiometry of this acid base reaction NaOH aq $\text{C}_6\text{H}_4\text{COOH}$ aq $\text{C}_6\text{H}_4\text{COONa}$ aq H_2O l the point of equivalence is the point of titration when the number of moles of naoh added is equal to the number of moles of KHP in the