

PROPERTIES OF MATTER STUDY GUIDE



ATOMS, ELEMENTS, COMPOUNDS + MIXTURES

Complete the vocabulary matching below by drawing a line to match the correct definition to the correct vocabulary word.

atom	d. A mixture in which the different parts can be seen because the substances are not evenly distributed.
element	b. The basic unit of matter that is made of protons, neutrons and electrons.
compound	c. A mixture in which the substances are evenly distributed and you cannot see the different parts.
homogeneous mixture	e. A pure substance that is made of 2 or more atoms that are chemically combined.
heterogeneous mixture	a. A pure substance that is made of only 1 type of atom and can be found on the periodic table.

Complete the chart below about the subatomic particles and their charges:

	PARTICLE #1	PARTICLE #2	PARTICLE #3
NAME	proton	neutron	electron
CHARGE	+ (positive)	0 (neutral)	- (negative)
LOCATION	part of nucleus	part of nucleus	outer "cloud-like" orbit

Complete the graphic organizer below about matter below. Need help? Use the word bank to assist you!

heterogeneous	element	mixture	compound
	pure substance	electron	homogeneous



Classify the following examples as elements, compounds or mixtures in the table below.

carbon (C)	salt	water (H ₂ O)
hydrogen (H)	salt water	sugar (sucrose)



ELEMENT	COMPOUND	MIXTURE
carbon (C) hydrogen (H)	water (H ₂ O) salt (NaCl) sugar (sucrose)	salad

Why is carbon considered an element and carbon dioxide considered a compound?

Carbon is an element on the periodic table, made of only carbon atoms. Carbon dioxide is carbon and oxygen chemically combined together.

Matter Review Guide Answer Key

Lingjun Ying



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Table of Contents Matter Review Guide Answer Key

1. Understanding the eBook Matter Review Guide Answer Key
 - The Rise of Digital Reading Matter Review Guide Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Matter Review Guide Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Matter Review Guide Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Matter Review Guide Answer Key
 - Personalized Recommendations
 - Matter Review Guide Answer Key User Reviews and Ratings
 - Matter Review Guide Answer Key and Bestseller Lists
5. Accessing Matter Review Guide Answer Key Free and Paid eBooks
 - Matter Review Guide Answer Key Public Domain eBooks
 - Matter Review Guide Answer Key eBook Subscription Services
 - Matter Review Guide Answer Key Budget-Friendly Options
6. Navigating Matter Review Guide Answer Key eBook Formats

- ePub, PDF, MOBI, and More
- Matter Review Guide Answer Key Compatibility with Devices
- Matter Review Guide Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Matter Review Guide Answer Key
 - Highlighting and Note-Taking Matter Review Guide Answer Key
 - Interactive Elements Matter Review Guide Answer Key
- 8. Staying Engaged with Matter Review Guide Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Matter Review Guide Answer Key
- 9. Balancing eBooks and Physical Books Matter Review Guide Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Matter Review Guide Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Matter Review Guide Answer Key
 - Setting Reading Goals Matter Review Guide Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Matter Review Guide Answer Key
 - Fact-Checking eBook Content of Matter Review Guide Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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