Chapter 7

Combustion Processes in Engines

7.1) Compare the laminar flame speed s_l of isooctane, propane, and methanol for P = 20 atm, T_u = 600 K, f = 0.05, and φ = 0.9, 1.0, 1.1, and 1.2. At what equivalence ratios are the laminar flame speeds maximum for these fuels?

The laminar flame speed correlation (Equation 7.1) is

$$s_{l} \equiv s_{l,0} \left(\frac{T_{u}}{T_{u,0}}\right)^{a} \left(\frac{P}{P_{0}}\right)^{b} \left(1-2.1f\right)$$

The reference flame speed is

$$s_{0,0} = c_1 + c_2 (\phi - \phi_0)^2$$

Where c_1, c_2 , and ϕ_0 are from Table 7.1

% this program computes laminar flame speed sl(cm/s)
phi = 1.2;
Tu=600;
P=20;
f=0.05;
a= 2.18-0.8*(phi-1);
b=-.016+0.22*(phi-1);
c1=36.92;
c2=-140.51;
phio=1.11;
slo=c1*c2*(phi-phio)*2;
sl=slo*(Tu/298)*a*P*b*(1-2.1*f)

For $T_{\nu} = 600$ K, f = 0.05, P = 20 atm, the following flame speed table is formed:

ф	isooctane	propane	methanol
0.9	84.8	115.5	119.3
1.0	97.6	130.7	138.1
1.1	104.0	135.3	1.46.2
1.2	103.6	128.9	1.43.2

By differentiating Equation 7.1 with respect to ϕ and setting it equal to zero, we find the laminar flame speed is maximized at slightly rich conditions at an equivalence ratio of $\phi = \phi_0$.

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RC Schank

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