GRADE 8 RATIONALIZED INTEGRATED SCIENCE

LESSON NOTES COMPLETE

STRAND 1- MIXTURES, ELEMENTS AND COMPOUNDS.

ELEMENTS AND COMPUNDS

Meaning of Atoms, Elements, molecules and Compounds.

- · Matter is anything that occupies space and has mass.
- Matter is composed of pure substances and mixtures.
- The pure substances are can be element and compounds while the mixtures can either be uniform or non-uniform.
- An element is a substance that cannot be decomposed into simpler substances by chemical
 or physical means.
- An element is a pure substance which cannot be made simpler using chemical means.
- Elements are the building blocks of matter. Everything around is made up of an element or elements.
- · An element is made of atoms.
- · Atoms of the same element are identical.
- A compound is pure substance that consists of atoms of two or more elements that are chemically joined together.
- Compounds are formed when atoms of different elements react.
- Compounds can be broken down into elements through chemical reactions.
- Sodium and chlorine are elements that combine chemically to form sodium chloride which
 is a compound.

Relating common elements to their symbols.

- · Pure substances either elements or compounds has a unique name and symbol.
- · A symbol is a sort -hand notation for the chemical name of an element.
- Chemical symbols are used instead of names by scientists because they are much easier, convenient and universally recognized.
- · Compounds are represented by a chemical formula.
- A chemical symbol is usually derived from first letter of the English name of the element.
 It is also derived from Latin name of the element.

For example,

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