

Experiment Problems 1

1. Synthesis reactions:



2. Synthesis by mixing solutions in dilute solutions:



3. Synthesis solubility product (K_{sp}) constants and equilibrium hydrolysis and acids:

	Elemental	Basic Difference
a)	Fe^{3+}	
b)		$2\text{Fe}^{3+} + 3\text{OH}^- \rightarrow$
c)	Cu^{2+}	

	Elemental	Basic Difference
a)	Fe^{3+}	
b)	Cu^{2+}	
c)		$\frac{1}{\text{OH}^-}$

4. Synthesis equilibrium constants: chemical constants are represented hydrolysis and acids:



Hs Mathematics Algebra 1 Tesccc 20excellent Exponents

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