Honors Chemistry

Electrochemistry Problems

Name	

Date Period

The voltnic cell shows above is composed of the metals Nickel (Ni) and Calcions (Ca) and their nimate salts. Assume both metals are 2" in their ionic state. Pay attention to the direction of electronflow. Answer the following questions using the letters indicated above. The same letter may be used more than once.

- Soln bridge
- Nickel electrode
- 3. Calcium nitrate solution
- Calcium electrode
- Nickel nitrate solution
- Anode:
- Cineits order.
- 8. Electrode where oxidation occurs
- Electrode where reduction occurs
- 11. Electrode that gets smaller with time
- 12. Solution that receives positive ions from the salt bridge



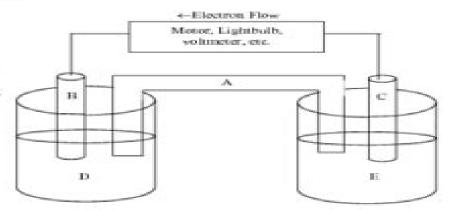
Write a condensed cell diagram for this reaction.

Write the oxidation half reaction for the cell.

Write the reduction half reaction for the cell.

For the following five questions identify the half reactions as either oxidized or reduced

which species is the redocing agent? Explain.



High School Electrochemistry Questions

Jicheng Xie

 ${\bf High\ School\ Electrochemistry\ Questions:}$

Reviewing **High School Electrochemistry Questions**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**High School Electrochemistry Questions**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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