

CHAPTER 17 STUDY GUIDE

Chemical Equilibrium

Section 17.1 Equilibrium: A State of Dynamic Balance

In your textbook, read about chemical equilibrium.

Complete each statement.

1. When a reaction results in almost complete conversion of reactants to products, chemists say the reaction goes to _____.
2. A reaction that can occur in both the forward and the reverse directions is called a(n) _____.
3. _____ is a state in which the forward and reverse reactions balance each other because they take place at equal rates.
4. At equilibrium, the concentrations of reactants and products are _____ but that does not mean that the amounts or concentrations are _____.
5. Equilibrium is a state of _____, not one of _____.

In your textbook, read about equilibrium expressions and constants.

For each statement below, write *true* or *false*.

- _____ 6. The law of chemical equilibrium states that at a given pressure, a chemical system may reach a state in which a particular ratio of reactant to product concentrations has a constant value.
- _____ 7. The equation $\text{H}_2(\text{g}) + \text{I}_2(\text{g}) \rightleftharpoons 2\text{HI}(\text{g})$ is an example of a homogeneous equilibrium.
- _____ 8. If an equilibrium constant has a value less than one, the reactants are favored at equilibrium.
- _____ 9. The value for K_{eq} is constant only at a specific volume.
- _____ 10. If the equilibrium constant for a reaction at 300 K is 49.7, the concentration of the reactants will be greater than the concentration of the products.
- _____ 11. A heterogeneous equilibrium means that reactants and products are present in more than one state.
- _____ 12. The product of the forward chemical reaction is HI, for the equilibrium expression:

$$K_{\text{eq}} = \frac{[\text{HI}]^2}{[\text{H}_2][\text{I}_2]}$$

Matter And Change Study Guide Answers

**John Krenos, Joseph Potenza, Loretta
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norm axis tight

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web feb 3 2004 `cdfplot x` displays a plot of the empirical cumulative distribution function cdf of the input array x in the current figure the empirical cdf $y = f(x)$ is defined as the proportion of x values less than or equal to x if input x is a matrix then `cdfplot x` parses it to the vector and displays cdf of all values

how to plot cdf in matlab tutorial youtube - Nov 06 2022

web feb 4 2021 this video describe the basics of a popular distribution and shows the coding simulation for cdf cumulative distribution function in matlab

cumulative distribution function matlab cdf mathworks - Dec 27 2021

web define the input vector x to contain the values at which to calculate the cdf `x = 2 1 0 1 2` compute the cdf values for the normal distribution with the mean μ equal to 1 and the standard deviation σ equal to 5 `mu = 1 sigma = 5 y = cdf normal x mu sigma`
`y = 1 5 0 2743 0 3446 0 4207 0 5000 0 5793`