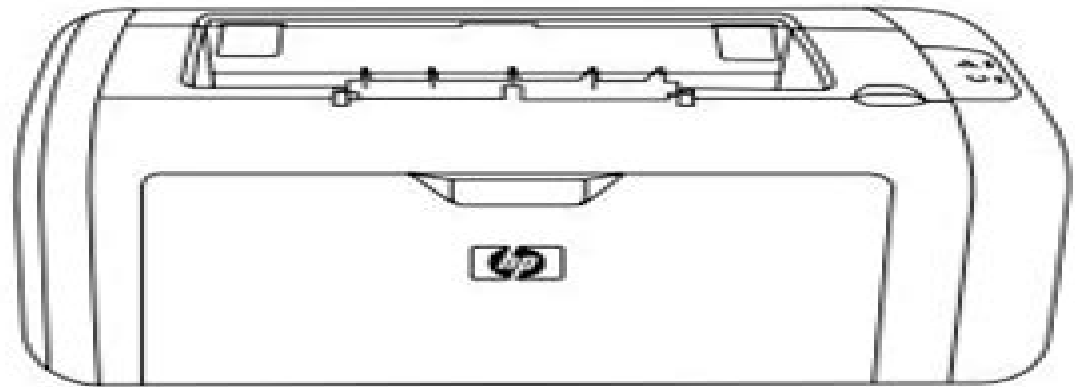


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$$\text{Pb(NO}_3)_2 \text{ (aq)} + 2 \text{HCl (aq)} \rightarrow \text{PbCl}_2 \text{ (s)} + 2 \text{HNO}_3$$

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web the net ionic equation for equation 1 is given as equation 3 $\text{s}^{2+}(\text{aq}) + \text{Fe}^{2+}(\text{aq}) \rightarrow \text{FeS}(\text{s})$ eq 3 8 note that the phase indicator aq is optional on ionic species but is

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