Advanced Hess's Law Practice Problems

 Given the following equations and △H^o values, determine the heat of reaction (kJ) at 298 K for the reaction:

$$B_2H_6(g) + 6 \text{ Cl}_2(g) \rightarrow 2 \text{ BCL}_2(g) + 6 \text{ HCl}(g)$$

 $G \circ O \cap \mathcal{F}$

 Given the following equations and AH^o values, determine the heat of reaction (kJ) at 298 K for the reaction:

20F2(9) + 25(5) -+ 502(9) + 5F4(9) AH=-319.5KJ

Hess S Law 12 3 Practice Problems

Julia Burdge

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