

Ionic Compounds: Names and Formulas Worksheet

1. Write the formulas for the following compounds.

- (a) sodium fluoride NaF
 (b) potassium chloride KCl
 (c) calcium oxide CaO
 (d) lithium bromide LiBr
 (e) beryllium chloride BeCl₂
 (f) potassium sulfide K₂S
 (g) magnesium nitride Mg₃N₂

- (h) calcium phosphide Ca₃P₂
 (i) magnesium oxide MgO
 (j) aluminum chloride AlCl₃
 (k) sodium iodide NaI
 (l) potassium nitride K₃N
 (m) lithium sulfide Li₂S
 (n) barium phosphide Ba₃P₂

2. Write the names for the following compounds.

- (a) LiCl lithium chloride
 (b) MgS Magnesium sulfide
 (c) NaCl sodium chloride
 (d) Al₂O₃ aluminum oxide
 (e) CaS calcium sulfide
 (f) Zn₃P₂ zinc phosphide
 (g) Be₃P₂ beryllium phosphide

- (h) Na₂O sodium oxide
 (i) KBr potassium bromide
 (j) BeO beryllium oxide
 (k) CaF₂ calcium fluoride
 (l) BaBr₂ barium bromide
 (m) Na₃N sodium nitride
 (n) Al₂S₃ aluminum sulfide

Naming Worksheet 5 Ionic Compounds Summary

Answer Key

Terry C. Jones



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