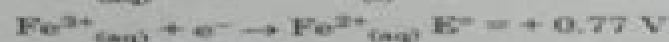
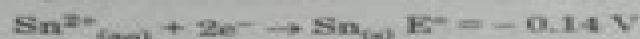


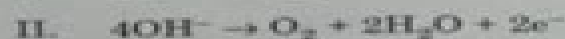
3. A voltaic cell is made by connecting two half cells represented by half equations below : 1



Which statement is correct about this voltaic cell ?

- (a) Fe^{2+} is oxidised and the voltage of the cell is -0.91 V
(b) Sn is oxidised and the voltage of the cell is 0.91 V
(c) Fe^{2+} is oxidised and the voltage of the cell is 0.91 V
(d) Sn is oxidised and the voltage of the cell is 0.63 V

4. Four half reactions I to IV are shown below : 1



Which two of these reactions are most likely to occur when concentrated brine is electrolysed ?

- (a) I and III (b) I and IV
(c) II and III (d) II and IV

5. Which property of transition metals enables them to behave as catalysts ? 1

- (a) High melting point (b) High ionisation enthalpy
(c) Alloy formation (d) Variable oxidation states

6. In the two tetrahedral structures of dichromate ion 1

- (a) 4 Cr - O bonds are equivalent in length.
(b) 6 Cr - O bonds are equivalent in length.
(c) All Cr - O bonds are equivalent in length.
(d) All Cr - O bonds are non-equivalent.



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