

Unit 6a Review questions HL (answers at end)

May 2019 HL, p2 T21

Brass is a copper-containing alloy with many uses. An analysis is carried out to determine the percentage of copper present in three identical samples of brass. The reactions involved in this analysis are shown below.



- (a) (i) Deduce the change in the oxidation numbers of copper and nitrogen in step 1. [2]

Copper:

.....

Nitrogen:

.....

- (ii) Identify the oxidising agent in step 1. [1]

.....

- (b) A student carried out this experiment three times, with three identical small brass nails, and obtained the following results.

Mass of brass = 0.456 g $\pm$ 0.001 g

Titre	1	2	3
Initial volume of 0.100 mol dm ⁻³ S ₂ O ₃ ²⁻ (± 0.05 cm ³)	0.00	0.00	0.00
Final volume of 0.100 mol dm ⁻³ S ₂ O ₃ ²⁻ (± 0.05 cm ³)	28.50	28.60	28.40
Volume added of 0.100 mol dm ⁻³ S ₂ O ₃ ²⁻ (± 0.10 cm ³)	28.50	28.60	28.40
Average volume added of 0.100 mol dm ⁻³ S ₂ O ₃ ²⁻ (± 0.10 cm ³)	28.50		

N4 Chemi Sp2 Eng Tz0 Xx

Yicheng Fang



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N4 Chemi Sp2 Eng Tz0 Xx Book Review: Unveiling the Power of Words

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Table of Contents N4 Chemi Sp2 Eng Tz0 Xx

1. Understanding the eBook N4 Chemi Sp2 Eng Tz0 Xx
 - The Rise of Digital Reading N4 Chemi Sp2 Eng Tz0 Xx
 - Advantages of eBooks Over Traditional Books
2. Identifying N4 Chemi Sp2 Eng Tz0 Xx
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an N4 Chemi Sp2 Eng Tz0 Xx
 - User-Friendly Interface
4. Exploring eBook Recommendations from N4 Chemi Sp2 Eng Tz0 Xx
 - Personalized Recommendations
 - N4 Chemi Sp2 Eng Tz0 Xx User Reviews and Ratings
 - N4 Chemi Sp2 Eng Tz0 Xx and Bestseller Lists

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 - N4 Chemi Sp2 Eng Tz0 Xx Public Domain eBooks
 - N4 Chemi Sp2 Eng Tz0 Xx eBook Subscription Services
 - N4 Chemi Sp2 Eng Tz0 Xx Budget-Friendly Options
6. Navigating N4 Chemi Sp2 Eng Tz0 Xx eBook Formats
 - ePub, PDF, MOBI, and More
 - N4 Chemi Sp2 Eng Tz0 Xx Compatibility with Devices
 - N4 Chemi Sp2 Eng Tz0 Xx Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of N4 Chemi Sp2 Eng Tz0 Xx
 - Highlighting and Note-Taking N4 Chemi Sp2 Eng Tz0 Xx
 - Interactive Elements N4 Chemi Sp2 Eng Tz0 Xx
8. Staying Engaged with N4 Chemi Sp2 Eng Tz0 Xx
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers N4 Chemi Sp2 Eng Tz0 Xx
9. Balancing eBooks and Physical Books N4 Chemi Sp2 Eng Tz0 Xx
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection N4 Chemi Sp2 Eng Tz0 Xx
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine N4 Chemi Sp2 Eng Tz0 Xx
 - Setting Reading Goals N4 Chemi Sp2 Eng Tz0 Xx
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of N4 Chemi Sp2 Eng Tz0 Xx
 - Fact-Checking eBook Content of N4 Chemi Sp2 Eng Tz0 Xx
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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