

Molecular Geometries of Covalent Molecules: Lewis Structures and the VSEPR Model

Pre-lab Questions

Before beginning this experiment in the laboratory, you should be able to answer the following questions.

1. Distinguish among ionic, covalent, and metallic bonding.
2. Which of the following molecules possess polar covalent bonds: H_2 , N_2 , HCl , HCN , and CO_2 ?
3. Which of the molecules in question 2 have molecular dipole moments?
4. What are the favored geometrical arrangements for AB_n molecules for which the A atom has 2, 3, 4, 5, and 6 pairs of electrons in its valence shell?
5. How many equivalent orbitals are involved in each of the following sets of hybrid orbitals: sp , sp^2 , sp^3d , and sp^3d^2 ?
6. Define the term *formal charge*.
7. Calculate the formal charges of all of the atoms in CO , CO_2 , and CO_3^{2-} .

Molecular Geometries Of Covalent Molecules Lab Answers

Terry C. Jones



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