

### Experiment 3: Chemical Kinetics

## Report Sheet/Post Lab Questions

Measure	Solution 1	Solution 2	Solution 3	† Temperature
Volume H <sub>2</sub> O <sub>2</sub> (mL)	5.0	10	5	5.0
Volume H <sub>2</sub> O (mL)	15.0	10.0	5.0	15.0
Volume KI (mL)	10	10	20	10
Slope	0.1484	0.2677	0.3129	0.3018

1. Show your calculation for the slope.

$$m = 1$$

$$n = 2$$

2. Write the Rate Law for this reaction

$$\text{rate} = k[A][B]^2$$

3. What is the order of this reaction?

$$m+n = 3$$

4. Did the rate for the higher temperature increase or decrease?

**higher**

5. Look up the solubility of oxygen in water.

**faster**

6. From the solubility value, do you expect that its solubility can affect the experimental results? Explain why.

7. What are some possible errors in this experiment?

**the time we get is no precise**

8. Was iodide left in the reaction solution after it was allowed to go to completion? What can you conclude about iodide?

9. From the data collected could the activation energy for the reaction be calculated?

# Kinetics Of A Reaction Lab Answers

**National Institute of Standards and  
Technology (U.S.), National Institute of  
Standards and Technology (U.S.).  
Information Resources and Services  
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## Kinetics Of A Reaction Lab Answers:

Reaction Rates for High-temperature Air with Carbon and Sodium Impurities Mina L. Carnicom, 1968 The values used by a number of investigators for the rate constants of high temperature greater than or equal to 1000 K homogeneous gaseous reactions involving species of the elements nitrogen oxygen carbon and sodium have been compiled and are presented in tabular form Included are reactions involving neutral species charged species free electrons some species in excited electronic or vibrational states and radiative processes      Scientific and Technical Aerospace Reports ,1986 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database      American Laboratory ,2003      Hydrogen Energy System Yuda Yürüm, 2012-12-06 In the near future the world will need to convert to a suitable clean energy supply one that will meet the demands of an increasing population while giving few environmental problems One such possible supply is hydrogen Hydrogen Energy System describes the present status of hydrogen as an energy supply as well as its prospect in the years to come It covers the transition to hydrogen based sustainable energy systems the technology of hydrogen production its storage and transport and current and future hydrogen utilisation Economic analyses of the hydrogen energy system together with case studies are also presented      **Report summaries** United States. Environmental Protection Agency, 1983      **Publications of the National Institute of Standards and Technology ... Catalog** National Institute of Standards and Technology (U.S.), National Institute of Standards and Technology (U.S.). Information Resources and Services Division, 1994      *Cracking the AP Chemistry Exam, 2009 Edition* Paul Foglino, 2009-01-01 Provides techniques for achieving high scores on the AP chemistry exam and includes two full length practice tests      **Cracking the AP Chemistry Exam, 2012 Edition** Paul Foglino, 2011-08-02 Provides techniques for achieving high scores on the AP chemistry exam and includes two full length practice tests      *Cracking the AP Chemistry Exam, 2013 Edition* Paul Foglino, Princeton Review, 2012-08-07 Provides techniques for achieving high scores on the AP chemistry exam and includes two full length practice tests a subject review for all topics and sample questions and answers      **Thermal Analysis** WIEDEMANN, 2013-11-11 Earlier efforts in the field of thermal analysis were concerned with the demonstration of the applicability of techniques to a broad spectrum of materials and to establish the relationship of such techniques with other more accepted method While such efforts will and should continue the Third International Conference was unique in that the first standards were disclosed for differential thermal analysis This was the culmination of the international cooperative effort of the ICTA s Standardization Committee The standards currently are available from the United State s National Bureau of Standards Thus thermal analysis can be considered to have attained its majority Realization of full maturity can be expected in the near future Inclusion of plenary lectures in these volumes represents a significant departure from previous Conferences This change is the result of the ICTA s recognition of its educational responsibilities In the Foreword of the

Proceedings of the Second International Conference Professor L Berg expressed the hope that thermal methods of analysis would find wider application in science and technology The citation above together with the papers presented indicate the fulfillment of this hope Xerox Corporation C B Murphy Rochester N Y U S A President ICTA 1968 1971 XIII PREFACE For the past two decades thermo analytical methods have reached a stage of considerable importance which is particularly due to the developments in the area of instrumentation Fossil Energy Update ,1982 **Enzymes and their Applications** Mr. Rohit Manglik,2024-05-17 EduGorilla Publication is a trusted name in the education sector committed to empowering learners with high quality study materials and resources Specializing in competitive exams and academic support EduGorilla provides comprehensive and well structured content tailored to meet the needs of students across various streams and levels

**Nuclear Science Abstracts** ,1974 **Chemistry in the Laboratory** James M. Postma,Julian L. Robert,J. Leland Hollenberg,2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations *Evaluated Kinetic Data for High Temperature Reactions* D. L. Baulch,1972 **Uranium • Plutonium Transplutonic Elements** Harold C. Hodge,J.N. Stannard,J.B. Hursh,2013-06-29 More than a year ago the three editors sat down at a table and worked out a set of six chapter headings which they believed might serve in turn for each of the three sections of this handbook The reader will note a similarity in order of presentation and in emphasis However as our editorial plans progressed it became apparent that for each element and for the element group there were one or two special topics appropriate for that section alone Accordingly in the section on uranium the common pattern holds for Chaps 1 through 6 which include an introduction Chap 1 a discussion of the physical and chemical properties Chap 2 experimental data on animals Chap 3 experimental data on man Chap 4 the rationale and development of air concentration limits to control industrial worker exposure Chap 5 and the practical problems of applying such limits in the uranium industry Chap 6 Chap 7 entitled Uranium Mining Hazards is the subject category which is special for uranium the chapter brings up to date the account of an important occupational hazard which was first noted by GEORGIUS AGRICOLA 1490 1555 *Energy Research Abstracts* ,1986 **Current Catalog** National Library of Medicine (U.S.),1968 First multi year cumulation covers six years 1965 70 **Selected Water Resources Abstracts** ,1976 **Advanced Chemistry** Michael Clugston,Rosalind Flemming,2000-06-08 Carefully researched by the authors to bring the subject of chemistry up to date this text provides complete coverage of the new A and AS level core specifications The inclusion of objectives and questions make it suitable for self study

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