#### **CHAPTER 20**

#### Accounting for Pensions and Postretirement Benefits

#### ASSIGNMENT CLASSIFICATION TABLE (BY TOPIC)

Topics		Questions	Brief Exercises	Exercises	Problems	Concepts for Analysis
	Basic definitions and concepts related to pension plans.	1, 2, 3, 4, 5, 6, 7, 8, 9, 12, 24, 30		16		1, 2, 3, 4, 5, 7
2.	Worksheet preparation.		3	3, 4, 7, 10, 14, 15, 18	1, 2, 4, 7, 8, 9, 10, 11, 12	
3.	Income statement recognition, computation of pension expense.	9, 10, 11, 13, 16, 17	1, 2, 4	1, 2, 3, 6, 11, 13, 14, 15, 16, 17, 18	1, 2, 3, 4, 5, 6, 9, 11, 12	4, 5
4.	Balance sheet recognition, computation of pension expense.	15, 19, 20, 22, 23	6, 10	3, 9, 11, 12, 13, 14	1, 2, 3, 4, 5, 6, 7, 8, 9, 11, 12	2, 5, 7
6,	Corridor calculation.	18	201	8, 13, 14, 16, 17, 18	2, 3, 5, 6, 7, 8, 11, 12	3, 4, 5, 6
G.	Prior service cost.	12, 13, 20	5, 6, 8	1, 2, 3, 5, 9, 11, 12, 13, 14	1, 2, 3, 4, 6, 7, 8, 9, 11, 12	1, 4
7.	Gains and losses.	14, 17, 21, 22	7. 9	8, 9, 13, 14, 16, 17	1, 2, 3, 4, 5, 6, 7, 8, 9, 11, 12	4, 5, 6
8.	Disclosure issues.	23	1.0	9, 11, 12	11, 12	3, 4
9.	Special Issues.	25				
*10.	Postretirement benefits.	26, 27, 28, 29	11, 12	19, 20, 21, 22, 23, 24	13, 14	

\*This material is dealt with in an Appendix to the chapter.

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for the nitration  $4\,01\,$  ml of nitric acid  $4\,00\,$  ml of sulfuric acid and  $4\,56\,$  ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only  $4\,56\,$  ml of bromobenzene was used the amount of product predicted to be recovered was  $8\,7533\,$  g

## lab 334 nitration of aromatic compounds preparation of studocu - Dec 23 2022

the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

experiment synthesis of 1 bromo 4 nitrobenzenee pbworks - May 28 2023

in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical x 100 what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer experiment 11 electrophilic aromatic substitution nitration - Jan 12 2022

on the right the arenium cation has a nitro set with incredibly small 0 6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group