Acids and Bases

How do acids and bases behave in water?

Why?

Acids and bases play an important role in our lives. Numerous biological processes, industrial applications, and even environmental problems are a function of the acidity or basicity (alkalinity) of aqueous solutions. It is therefore important to understand what makes a substance behave as an acid or a base when dissolved in water. In this activity, we will explore the physical and chemical properties of acids and bases.

Model 1 – Arrhenius Acids and Bases

Common Name for Aqueous Solution	Chemical Formula	Found in	Tastes	Turns Litmus Paper	Conducts Electricity?	Acid or Base
Acetic acid	HC,H,O,(aq)	Vinegar	Sour	Red	Yes	Acid
Benzoic acid	HC,H,O,(aq)	Food preservative	Sour	Red	Yes	Acid
Phosphoric acid	H,PO4(aq)	Soda pop	Sour	Red	Yes	Acid
Hydrochloric acid	HCl(aq)	Stomach acid	Sour	Red	Yes	Acid
Citric acid	H ₄ C ₄ H ₄ O ₅ (aq)	Citrus fruits	Sour	Red	Yes	Acid
Ascorbic acid	$H_2C_aH_aO_a(aq)$	Vitamin C	Sour	Red	Yes	Acid
Magnesium hydroxide	Mg(OH),(aq)	Milk of magnesia	Bitter	Blue	Yes	Base
Aluminum hydroxide	Al(OH),(aq)	Antacids	Bitter	Blue	Yes	Basse
Barium hydroxide	Ba(OH),(aq)	Lubricants	POISON	Blue	Yes	Base
Sodium hydroxide	NaOH(aq)	Drain cleaner	POISON	Blue	Yes	Base

- 1. Refer to Model 1.
 - What is the common chemical name for Vitamin C? ascorbic acid
 - b. Is Vitamin C classified as an acid or a base? acid
- Examine the properties of the Arrhenius acids in Model 1. List three properties that all Arrhenius acids have in common.

sour, red, conduct electricity

- 3. Examine the chemical formulas for the Arrhenius acids in Model 1. What feature do all the Arrhenius acid chemical formulas have in common?
- Examine the properties of the Arrhenius bases in Model 1. List two properties that all Arrhenius bases have in common.

Introduction To Acids And Bases Answers Pogil

RJ Alexander

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POGIL Shawn R. Simonson, 2023-07-03 Process Oriented Guided Inquiry Learning POGIL is a pedagogy that is based on research on how people learn and has been shown to lead to better student outcomes in many contexts and in a variety of academic disciplines Beyond facilitating students mastery of a discipline it promotes vital educational outcomes such as communication skills and critical thinking Its active international community of practitioners provides accessible educational development and support for anyone developing related courses Having started as a process developed by a group of chemistry professors focused on helping their students better grasp the concepts of general chemistry The POGIL Project has grown into a dynamic organization of committed instructors who help each other transform classrooms and improve student success develop curricular materials to assist this process conduct research expanding what is known about learning and teaching and provide professional development and collegiality from elementary teachers to college professors As a pedagogy it has been shown to be effective in a variety of content areas and at different educational levels This is an introduction to the process and the community Every POGIL classroom is different and is a reflection of the uniqueness of the particular context the institution department physical space student body and instructor but follows a common structure in which students work cooperatively in self managed small groups of three or four The group work is focused on activities that are carefully designed and scaffolded to enable students to develop important concepts or to deepen and refine their understanding of those ideas or concepts for themselves based entirely on data provided in class not on prior reading of the textbook or other introduction to the topic The learning environment is structured to support the development of process skills such as teamwork effective communication information processing problem solving and critical thinking The instructor s role is to facilitate the development of student concepts and process skills not to simply deliver content to the students The first part of this book introduces the theoretical and philosophical foundations of POGIL pedagogy and summarizes the literature demonstrating its efficacy The second part of the book focusses on implementing POGIL covering the formation and effective management of student teams offering guidance on the selection and writing of POGIL activities as well as on facilitation teaching large classes and assessment The book concludes with examples of implementation in STEM and non STEM disciplines as well as guidance on how to get started Appendices provide additional resources and information about Analytical Chemistry Juliette Lantz, Renée Cole, The POGIL Project, 2014-12-31 An essential guide to The POGIL Project inquiry approach instrumental analysis Analytical Chemistry offers an essential guide to inquiry approach instrumental analysis collection The book focuses on more in depth coverage and information about an inquiry approach This authoritative guide reviews the basic principles and techniques Topics covered include method of standard the microscopic view of electrochemistry calculating cell potentials the BerriLambert atomic and molecular absorption processes vibrational modes mass spectra interpretation and much more Acids and Bases Kristi Lew, 2009 Learn about acids and bases chemical

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structure and acid base properties of mixed metal oxides and correlation between acid base properties and catalytic activity and selectivity are also deliberated This publication is useful to professional chemists and graduate students in the fields of organic inorganic and physical chemistry petroleum chemistry and catalysis including readers interested in the acidic and basic properties on solid surfaces **Introduction to Chemical Principles** Edward I. Peters, 1978 Chemistry and Acid-base Properties of Surfaces Jean-Charles Joud, Marie-Geneviève Barthés-Labrousse, 2015 Acids Acids and Bases Ray T. Slattery, 1975 The Electronic Theory of Acids and and Bases Ronald Percy Bell,1952 Acids and Bases Carol Baldwin, 2004 Strong Acids and **Bases** William Fay Luder, Saverio Zuffanti, 1946 Bases-General Properties ,2016 You return to the topic of pH and learn about how pH relates to two kinds of compounds acids and bases Through an introductory problem you explore the relationship of various ions within these compounds **Science** ,2017

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