

Ch. 4 Guided Reading

4.1 Introduction to Atoms

Guided Reading and Study

Use Target Reading Skills

Sample questions and answers:

Q. What particles are in the center of an atom?

A. Protons and neutrons.

Q. What particles move around the outside of the nucleus?

A. Electrons.

1. False

2. a, c, d

3. True

4. J. J. Thomson

5. gold-foil experiment

6. Protons

7. Niels Bohr

8. Neutron

9. False

10. a, b

11. a. nucleus

b. neutron

c. proton

d. electron cloud

12. The number of protons in an atom equals

the number of electrons. The positive charge from the protons balances the negative charge from the electrons, making the atom neutral.

13. protons and neutrons

14. The mass of an electron is about 1/2,000 (Or 1/1,836) of the mass of a proton.

15. protons

16. The atomic number is the number of protons in the nucleus.

17. Isotopes are atoms with the same number of protons but with a different number of neutrons.

18. Student diagrams should indicate any two of the following: An isotope of carbon with a mass number of 12 has 6 protons, 6 neutrons, and 6 electrons. An isotope of carbon with a mass number of 13 has 6 protons, 7 neutrons, and 6 electrons. An isotope of carbon with a mass number of 14 has 6 protons, 8 neutrons, and 6 electrons.

4.2 Organizing the Elements

Guided Reading and Study

1. He recognized a set of patterns that applied to all elements.

Introduction To Atoms Guided Reading And Study Answers

Zhe Xu



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Life Study Guide David E. Sadava, Gordon H. Orians, Craig Heller, William K. Purves, 2006-12-22 Especially helpful for AP Biology students each chapter of the study guide offers a variety of study and review tools The contents of each chapter are broken down into both a detailed review of the Important Concepts covered and a boiled down Big Picture snapshot The guide also covers study strategies common problem areas and provides a set of study questions both multiple choice and short answer

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Life: The Science of Biology Study Guide William K. Purves, Edward Dzialowski, Lindsay Goodloe, Betty McGuire, Nancy Guild, Paula Mabee, 2003-12-26 New edition of a text presenting underlying concepts and showing their relevance to medical agricultural and environmental issues Seven chapters discuss the cell information and heredity evolutionary process the evolution of diversity the biology of flowering plants and of animals and ecology and biogeography Topics are linked by themes such as evolution the experimental foundations of knowledge the flow of energy in the living world the application and influence of molecular techniques and human health considerations Includes a CD ROM which covers some of the subject matter and introduces and illustrates 1 700 plus key terms and concepts Annotation copyrighted by Book News Inc Portland OR

Study Guide, Introductory Chemistry, a Foundation ; Introductory Chemistry ; Basic Chemistry Iris Stovall, Donald J. DeCoste, 2000

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Science , 2004 **Resources in Education** , 1996-10 **Addison-Wesley Chemistry** Antony C. Wilbraham, Dennis D.

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one another but brought together in a more productive theoretical and practical relationship While the potential importance of this project is clear Laruelle remains famously difficult Anthony Paul Smith provides an introduction and guide to the text that situates you amongst the figures and concepts Laruelle engaged with provides a foothold for your own understanding and more importantly potential use of the project of non philosophy

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