

Name: _____ Date: _____ Period: _____

NOVA: Hunting the Elements

Part 1: Basic Chemistry

Blank Periodic Table

Fill in the parts of this periodic table covered by the documentary.

1 H																		2 He																			
3 Li		4 Be																		5 B		6 C		7 N		8 O		9 F		10 Ne							
11 Na		12 Mg																		13 Al		14 Si		15 P		16 S		17 Cl		18 Ar							
19 K		20 Ca		21 Sc		22 Ti		23 V		24 Cr		25 Mn		26 Fe		27 Co		28 Ni		29 Cu		30 Zn		31 Ga		32 Ge		33 As		34 Se		35 Br		36 Kr			
37 Rb		38 Sr		39 Y		40 Zr		41 Nb		42 Mo		43 Tc		44 Ru		45 Rh		46 Pd		47 Ag		48 Cd		49 In		50 Sn		51 Sb		52 Te		53 I		54 Xe			
55 Cs		56 Ba		57-78 *		71 Lu		72 Hf		73 Ta		74 W		75 Re		76 Os		77 Ir		78 Pt		79 Au		80 Hg		81 Tl		82 Pb		83 Bi		84 Po		85 At		86 Rn	
87 Fr		88 Ra		89-102 * (*) (*)		103 Lr		104 Rf		105 Db		106 Sg		107 Bh		108 Hs		109 Mt		110 Uun		111 Uuu		112 Uub		114 Uuq											

¹⁰ *Journal of the American Academy of Religion*, 47 (1979), 1-27.

57 La Lanthanum	58 Ce Cerium	59 Pr Praseodymium	60 Nd Neodymium	61 Pm Promethium	62 Sm Samarium	63 Eu Europium	64 Gd Gadolinium	65 Tb Terbium	66 Dy Dysprosium	67 Ho Holmium	68 Er Erbium	69 Tm Thulium	70 Yb Ytterbium
89 Ac Actinium	90 Th Thorium	91 Pa Protactinium	92 U Uranium	93 Np Neptunium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	97 Bk Berkelium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium

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Gold - Au

1. Write the number of subatomic particles in gold:
 - a. Protons:
 - b. Neutrons:
 - c. Electrons:
2. Give one property of gold.
3. How much gold is in one ton of the mined rock?

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