

# Chapter Test Chapter 10: Matter and Energy

## Matter and Energy

Write each answer in the space provided on the paper at the top or bottom of each multiple-choice question or in the answer line for each free-response question.

1. ☐ 1. The quantity of matter contained in a system is known as the system's  
a. volume  
b. mass  
c. density  
d. temperature
2. ☐ 2. According to the law of conservation of mass,  
a. the combined mass of the products is greater than the combined mass of the reactants.  
b. the combined mass of the products is less than the combined mass of the reactants.  
c. the combined mass of the products is equal to the combined mass of the reactants.  
d. matter is not conserved during a chemical reaction.
3. ☐ 3. According to the law of conservation of energy, energy is neither created nor destroyed.  
a. energy is created, but mass is conserved in a reaction.  
b. energy is conserved, but mass is not conserved in a reaction.  
c. energy is conserved, but mass is not conserved in a reaction.  
d. energy is conserved, but mass is not conserved in a reaction.
4. ☐ 4. Which of the following statements about energy is true?  
a. Energy is conserved, but mass is not conserved.  
b. Energy is conserved, but mass is not conserved.  
c. Energy is conserved, but mass is not conserved.  
d. Energy is conserved, but mass is not conserved.
5. ☐ 5. Which of the following is not a unit of energy?  
a. joule  
b. calorie  
c. kilocalorie  
d. watt
6. ☐ 6. The quantity of matter in a system is known as the system's  
a. mass  
b. volume  
c. density  
d. temperature
7. ☐ 7. The quantity of matter in a system is known as the system's  
a. mass  
b. volume  
c. density  
d. temperature

# Holt Chemistry Test Solutions Answers

**A Loxley**



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