

Holt Chemistry – Guided Notes, Chapter 1: The Science of Chemistry
Section 1: What is Chemistry?

Objectives:

- **Describe** ways in which chemistry is a part of your daily life.
- **Describe** the characteristics of three common states of matter.
- **Describe** physical and chemical changes, and give examples of each.
- **Identify** the reactants and products in a chemical reaction.
- **List** four observations that suggest a chemical change has occurred.

A _____ is any substance that has a definite composition. A _____ is the process by which one or more substances change to produce one or more new substances. The _____ of _____ are the physical forms of matter which are solid, liquid, gas, and plasma.

- _____ have a fixed volume and shape that result from the way their particles are arranged.
- _____ have a fixed volume but not a fixed shape.
- _____ have neither fixed volume nor shape.



Physical changes are changes in which the _____ of a substance doesn't change.

- Changes of state are physical changes.

Chemical changes occur when the _____ of substances change and new substances form.

Evidence of a chemical change:

			
a When acetic acid, vinegar, and sodium hydrogen carbonate, or baking soda, are mixed, the solution bubbles as carbon dioxide forms.	b When solutions of sodium sulfide and cadmium nitrate are mixed, cadmium sulfide, a solid precipitate, forms.	c When aluminum reacts with iron(III) oxide in the clay pot, energy is released as heat and light.	d When phenolphthalein is added to ammonia dissolved in water, a color change from colorless to pink occurs.

Holt Chemistry Section One Assessment Quiz Answers

Xiufeng Liu, William J. Boone

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science education It includes a conceptual introduction chapter and a set of individual chapters The introductory chapter reviews published studies applying Rasch measurement in the field of science education and identify important principles of Rasch measurement and best practices in applications of Rasch measurement in science education The individual chapters contributed by authors from Canada China Germany Philippines and the USA cover a variety of current topics on measurement concerning science conceptual understanding scientific argumentation scientific reasoning three dimensional learning knowledge in use and cross cutting concepts of the Next Generation Science Standards medical education learning experiences machine scoring bias formative assessment and teacher knowledge of argument There are additional chapters on advances in Rasch analysis techniques and technology including R Bayesian estimation comparison between joint maximum likelihood JML and marginal maximum likelihood MML estimations on model data fit and enhancement to Rasch models by Cognitive Diagnostic Models and Latent Class Analysis The volume provides readers who are new and experienced in applying Rasch measurement with advanced and exemplary applications in the forefront of various areas of science education research

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