

Chapter Test Chapter 10: Matter and Energy

Matter and Energy

Write each answer in the space provided on the paper at the back of this notebook. Show all work for problems that have multiple steps.

1. ☐ 1. The equation below defines potential energy.

 - a. $PE = mgh$
 - b. $PE = mv^2$
 - c. $PE = mgh^2$
 - d. $PE = mgh$

2. ☐ 2. Kinetic energy is defined as

 - a. the energy of motion.
 - b. the energy of position.
 - c. the energy of mass.
 - d. the energy of velocity.

3. ☐ 3. Kinetic energy is defined as the energy of motion.

 - a. $KE = \frac{1}{2}mv^2$
 - b. $KE = mv^2$
 - c. $KE = \frac{1}{2}mv$
 - d. $KE = mv$

4. ☐ 4. Which of the following is not a form of energy?

 - a. heat
 - b. light
 - c. mass
 - d. sound

5. ☐ 5. Which of the following is not a form of energy?

 - a. heat
 - b. light
 - c. mass
 - d. sound

6. ☐ 6. The equation below defines potential energy.

 - a. $PE = mgh$
 - b. $PE = mv^2$
 - c. $PE = mgh^2$
 - d. $PE = mgh$

7. ☐ 7. Kinetic energy is defined as the energy of motion.

 - a. $KE = \frac{1}{2}mv^2$
 - b. $KE = mv^2$
 - c. $KE = \frac{1}{2}mv$
 - d. $KE = mv$

Holt Chemistry Chapter 2test

Samuel Ralph Powers



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Holt Chemistry Ralph Thomas Myers, 2004 **KY HS Test Prac Wkbks W/Corr Sci 2001** Holt Rinehart & Winston, 2002-03 *Principles of Modern Chemistry* David W. Oxtoby, H. Pat Gillis, Laurie J. Butler, 2016-01-01 Long considered the standard for honors and high level mainstream general chemistry courses PRINCIPLES OF MODERN CHEMISTRY continues to set the standard as the most modern rigorous and chemically and mathematically accurate text on the market This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure Chapter 6 Electrochemistry Chapter 17 and Molecular Spectroscopy and Photochemistry Chapter 20 In addition the text utilizes mathematically accurate and artistic atomic and molecular orbital art and is student friendly without compromising its rigor End of chapter study aids focus on only the most important key objectives equations and concepts making it easier for students to locate chapter content while applications to a wide range of disciplines such as biology chemical engineering biochemistry and medicine deepen students understanding of the relevance of chemistry beyond the classroom **Laboratory Experiments to Accompany "Modern Chemistry,"** Fredus Nelson Peters, 1902

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