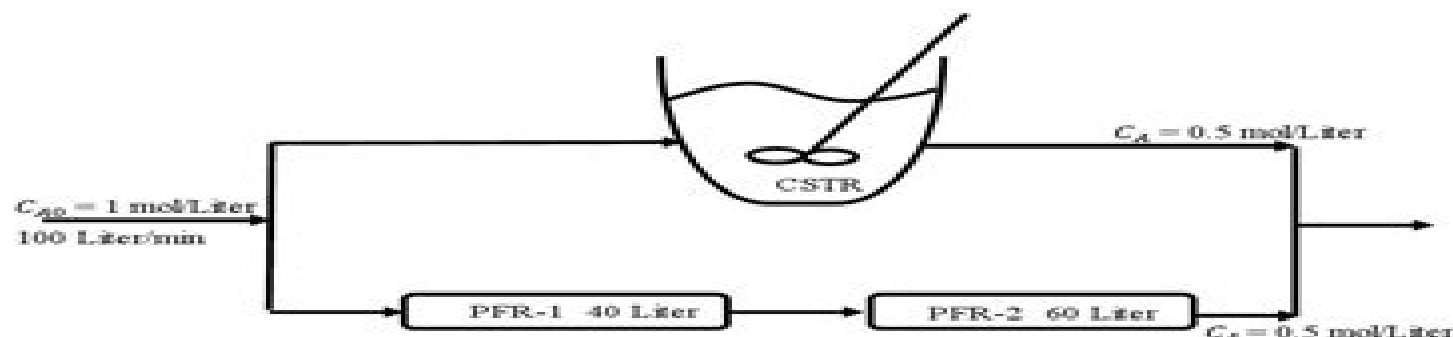


- Q.46 A first-order irreversible liquid phase reaction $A \rightarrow B$ ($k = 0.1 \text{ min}^{-1}$) is carried out under isothermal, steady state conditions in the following reactor arrangement comprising an ideal CSTR (Continuous-Stirred Tank Reactor) and two ideal PFRs (Plug Flow Reactors). From the information in the figure, the volume of the CSTR (in Liters) is _____ (rounded off to the nearest integer).



- Q.47 The elementary liquid-phase irreversible reactions



take place in an isothermal ideal CSTR (Continuous-Stirred Tank Reactor). Pure A is fed to the reactor at a concentration of 2 mol/Liter. For the residence time that maximizes the exit concentration of B, the percentage yield of B, defined as $\left(\frac{\text{net formation rate of B}}{\text{consumption rate of A}} \times 100 \right)$, is _____ (rounded off to the nearest integer).

- Q.48 The elementary irreversible gas-phase reaction $A \rightarrow B + C$ is carried out adiabatically in an ideal CSTR (Continuous-Stirred Tank Reactor) operating at 10 atm. Pure A enters the CSTR at a flow rate of 10 mol/s and a temperature of 450 K. Assume A, B and C to be ideal gases. The specific heat capacity at constant pressure ($C_{p,i}$) and heat of formation ($H_{f,i}^\circ$), of component i ($i = A, B, C$), are:

$$\begin{array}{lll} C_{pA} = 30 \text{ J/(mol K)} & C_{pB} = 10 \text{ J/(mol K)} & C_{pC} = 20 \text{ J/(mol K)} \\ H_A^\circ = -90 \text{ kJ/mol} & H_B^\circ = -54 \text{ kJ/mol} & H_C^\circ = -45 \text{ kJ/mol} \end{array}$$

The reaction rate constant k (per second) $= 0.133 \exp \left\{ \frac{E}{R} \left(\frac{1}{450} - \frac{1}{T} \right) \right\}$, where $E = 31.4 \text{ kJ/mol}$ and universal gas constant $R = 0.082 \text{ L atm/(mol K)} = 8.314 \text{ J/(mol K)}$. The shaft work may be neglected in the analysis, and specific heat capacities do not vary with temperature. All heats of formation are referenced to 273 K. The reactor volume (in Liters) for 75% conversion is _____ (rounded off to the nearest integer).

Graduate Aptitude Test In Engineering Gate 20question Paper

Lingjun Ying



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Managing India R Rajesh Babu, Manish Thakur, 2024-03-12 This book explores the history and metamorphosis of the Indian Institutes of Management IIMs the premier business and management schools in India and their significance within the changing landscape of higher education nation building and socio economic development in the country Over the past decades IIMs as institutions have recalibrated their goals and priorities to address

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Chemistry in Daily Life is a comprehensive textbook designed to connect fundamental chemical principles with everyday applications In this book have 23 chapter each chapter explain with suitable figure and in last section of each chapter have multiple type question with answer This book explores how chemistry influences food environment health and industry It covers the composition and quality analysis of dairy products beverages food additives including artificial sweeteners Flavors preservatives colorants and contaminants Chapters paints dyes pigments and their industrial applications Air and water pollution are observed finished topics like air pollutants greenhouse gases acid rain and wastewater treatment Readers gain understanding into water purification methods the hydrologic cycle and soil chemistry The role of fertilizers and their classifications is explained alongside the chemistry of essential biomolecules like vitamins carbohydrates oils Medicinal chemistry is detailed discussions on analgesics antibiotics antimalarials other drugs The book also introduces concepts in chemical kinetics thermodynamics and energy systems including fuel cells batteries solar energy Finally environmental challenges such as plastic waste heavy metal pollution are addressed with eco friendly solutions Rich in real life examples diagrams exercises this book is perfect for students educators and anyone interested how chemistry shapes our world

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