

QUESTION 2

	Test	Observation	Inference
(a)(i)	C + Conc. H_2SO_4 + strong heat	A copious brown irritating fume which turned moist blue litmus paper to red	Gas is NO_2 NO_3^- present
(b)(i)	C + water + NaOH in drops, in excess	white gelatinous precipitate precipitate dissolved	Al^{3+} or Zn^{2+}
(ii)	C + water + NH_3 in drops, in excess	white gelatinous precipitate precipitate dissolved	Al^{3+} or Zn^{2+}
(iii)	C + water + freshly prepared $FeSO_4$ + Conc. H_2SO_4	A brown ring is formed at the junction of the acid and aqueous solution.	NO_3^-
(c)(i)	D + water	D dissolved	soluble
(ii)	Solution D + dil HCl	Colourless gas, smells like rotten egg, turns moist blue litmus paper to red and lead (II) ethanoate paper to black	H_2S present
(iii)	Solution D + NaOH in drops, in excess	white precipitate precipitate insoluble	Ca^{2+} or Pb^{2+}
(iv)	solution D + $(NH_4)_2CO_3$	white precipitate	Ca^{2+}

Test tube slightly warm

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SA Dillow



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