

Naming Ionic Compounds – Answer Key

Give the name of the following ionic compounds:

		Name
1)	Na_2CO_3	sodium carbonate
2)	NaOH	sodium hydroxide
3)	MgBr_2	magnesium bromide
4)	KCl	potassium chloride
5)	FeCl_2	iron (II) chloride
6)	FeCl_3	iron (III) chloride
7)	Zn(OH)_2	zinc hydroxide
8)	Be_2SO_4	beryllium sulfate
9)	CrF_2	chromium (II) fluoride
10)	Al_2S_3	aluminum sulfide
11)	PbO	lead (II) oxide
12)	Li_3PO_4	lithium phosphate
13)	TiI_4	titanium (IV) iodide
14)	Co_3N_2	cobalt (II) nitride
15)	Mg_3P_2	magnesium phosphide
16)	$\text{Ga(NO}_2)_3$	gallium nitrite
17)	Ag_2SO_3	silver sulfite
18)	NH_4OH	ammonium hydroxide
19)	Al(CN)_3	aluminum cyanide
20)	$\text{Be(CH}_3\text{COO)}_2$	beryllium acetate

Naming Ionic Compounds Key

H Kauffman

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Naming Ionic Compounds Key:

Naming Ionic Compounds Key Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Naming Ionic Compounds Key**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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Naming Ionic Compounds Key Introduction

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