

Name: _____ Date: _____ Lab Section _____

Pre-lab Study Questions:

1. Objectives of the experiment (in your own words):
2. What kinds of electrons form the covalent bonds in molecules?
3. Is a polar molecule symmetrical or asymmetrical?

FORMULA OF MOLECULE	LEWIS STRUCTURE	3-D SHAPE w/lone pairs	MOLECULAR GEOMETRY	MOLECULAR POLARITY
example: CH_4			tetrahedral	nonpolar
H_2O			tetrahedral	polar
H_2O^+			tetrahedral	nonpolar
C_2H_6				
C_2H_4			trigonal planar	Non polar
C_2H_2				
CO_2			Linear	Non polar
H_2O_2				
CH_3Cl				

Molecular Model Laboratory Manual Answers

Sabine Zange



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Reviews in Computational Chemistry, Volume 4 Kenny B. Lipkowitz,Donald B. Boyd,2009-09-22 This volume in the series brings together reknowned experts in the field to present the reader with an account of the latest developments in quantum mechanics molecular dynamics and the teaching of computational chemistry There are so many developments in the field of computational chemistry that it is difficult to keep track of them The series was established to review the high volume of developments in the field Rather than create a traditional article each author approaches a topic to enable the reader to understand and solve problems and locate key references quickly Each article has tutorial value An updated compendium of software for molecular modeling appears as an appendix as in previous volumes To the editors knowledge this is the most complete listing of sources of software for computational chemistry anywhere

Course and Curriculum Improvement Projects: Mathematics, Science, Social Sciences National Science Foundation (U.S.),1966

Laboratory Manual of Physical Chemistry Albert Watson Davison, Henry Sjoerd van Klooster,1922

Annual Report National Institutes of Health (U.S.). Division of Computer Research and Technology,1989

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