# **Mole Calculations**

#### Multiple Choice Questions

1

How many moles are there in 6 dm3 of oxygen at r.t.p.?

A 0.25 moles

B 0.50 moles

C 0.60 moles

D 3.00 moles

2

What is the total volume of gas remaining after 20 cm<sup>3</sup> ethane are burned completely in 100 cm3 oxygen? All volumes are measured at the same pressure and the same temperature, which is above 100 °C.

C<sub>2</sub>H<sub>6</sub> + 3 ½ O<sub>2</sub> → 2CO<sub>2</sub> + 3H<sub>2</sub>O

A 40 cm<sup>3</sup>

B 100 cm<sup>3</sup>

C 120 cm<sup>3</sup>

D 130 cm<sup>3</sup>

3

What is the mass of 0.36 moles of ethanoic acid, CH<sub>3</sub>COOH?

A 0.06g 🔘

B 21.60g (

C 6.00g  $\bigcirc$ 

D 36.00g (

# **Mole Practice Problems**

Lisa A. Seidman, Cynthia J. Moore, Jeanette Mowery

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